

6. Write a pair of balanced half-reaction equations—one showing a gain of electrons and one showing a loss of electrons—for each of the following reactions:



9. For each of the following, write and label the oxidation and reduction half-reaction equations. Ignore spectator ions.



10. We have only looked at one type of single displacement reaction—a metal displacing another metal from an ionic compound. A nonmetal can also displace another nonmetal from an ionic compound. For example,



Using your knowledge of atoms and ions and the ideas presented in this chapter, write a pair of balanced half-reaction equations for this reaction—one showing a gain of electrons and one showing a loss of electrons.



Figure 9
The reduction of iron(III) oxide by aluminum is called the “thermite” reaction. Observe this reaction in

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Tejal Desai, Sangeeta N. Bhatia



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Table of Contents Self Quiz Chapter Nelson Chemistry 1

1. Understanding the eBook Self Quiz Chapter Nelson Chemistry 1
 - The Rise of Digital Reading Self Quiz Chapter Nelson Chemistry 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Self Quiz Chapter Nelson Chemistry 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Self Quiz Chapter Nelson Chemistry 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Self Quiz Chapter Nelson Chemistry 1
 - Personalized Recommendations
 - Self Quiz Chapter Nelson Chemistry 1 User Reviews and Ratings

- Self Quiz Chapter Nelson Chemistry 1 and Bestseller Lists
- 5. Accessing Self Quiz Chapter Nelson Chemistry 1 Free and Paid eBooks
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 - Self Quiz Chapter Nelson Chemistry 1 eBook Subscription Services
 - Self Quiz Chapter Nelson Chemistry 1 Budget-Friendly Options
- 6. Navigating Self Quiz Chapter Nelson Chemistry 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Self Quiz Chapter Nelson Chemistry 1 Compatibility with Devices
 - Self Quiz Chapter Nelson Chemistry 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Self Quiz Chapter Nelson Chemistry 1
 - Highlighting and Note-Taking Self Quiz Chapter Nelson Chemistry 1
 - Interactive Elements Self Quiz Chapter Nelson Chemistry 1
- 8. Staying Engaged with Self Quiz Chapter Nelson Chemistry 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Self Quiz Chapter Nelson Chemistry 1
- 9. Balancing eBooks and Physical Books Self Quiz Chapter Nelson Chemistry 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Self Quiz Chapter Nelson Chemistry 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Self Quiz Chapter Nelson Chemistry 1
 - Setting Reading Goals Self Quiz Chapter Nelson Chemistry 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Self Quiz Chapter Nelson Chemistry 1
 - Fact-Checking eBook Content of Self Quiz Chapter Nelson Chemistry 1
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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