

# PERIODIC TABLE (ADVANCED)

## Molecular Structure of Hydrogen



### Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point:  $-259.14^{\circ}\text{C}$  ( $14.009185^{\circ}\text{K}$ ,  $-434.45203^{\circ}\text{F}$ )

Boiling Point:  $-252.87^{\circ}\text{C}$  ( $20.280005^{\circ}\text{K}$ ,  $-423.166^{\circ}\text{F}$ )

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $0.000089\text{ g/cm}^3$

Color: colorless

### Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

### Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

### Facts

Name Origin: from words *hydro* (water) and *genes* (generated)

Uses: balloons, metal refining

Obtained From: mines, oil, gas wells

## Molecular Structure of Carbon



Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point:  $3500.0^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6332.0^{\circ}\text{F}$ )

Boiling Point:  $4827.0^{\circ}\text{C}$  ( $5100.15^{\circ}\text{K}$ ,  $8700.6^{\circ}\text{F}$ )

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $2.62\text{ g/cm}^3$

Color: May be black

### Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

### Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

### Facts

Uses: steel, fibers

Obtained From: burning with insufficient oxygen

# Periodic Table Advanced Speedy Study Guide

**Ana Hategan, James A. Bourgeois, Tracy  
Cheng, Julie Young**



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