

## Chapter 6 Worksheet 1 - Key

Name \_\_\_\_\_

Period \_\_\_\_\_

1. Who is credited with developing the first periodic table?  
**Mendeleev**
2. How was the first periodic table organized?  
**Elements were placed in order of increasing atomic mass into columns with similar properties.**
3. What was the major problem with the periodic table developed by Mendeleev?  
**Some elements had to be placed out of order according to their atomic masses so that their properties would line up with the groups they were placed in.**
4. How did Henry Moseley update the periodic table?  
**Moseley organized elements by increasing atomic number.**
5. What is the group and period number within the periodic table for the element silver ( $Z = 47$ )? **Group 11 Period 5**
6. What is the octet rule?  
**The octet rule states that an atom with eight electrons in its outer level is stable.**
7. Explain why helium is very stable even though it does not have an octet.  
**Helium only has one energy level. This first energy level only requires two electrons to fill it instead of the normal eight electrons.**

For questions 8 through 17 give the family name that is being described.

8. Made up of the elements in group 18. **Noble Gases**
9. Some of these metals can float on water. **Alkali Metals**
10. Less reactive harder metals. **Transition Metals**
11. The most reactive of the nonmetals. **Halogens**
12. Elements in group 16. **Chalcogens**
13. All of these elements are found in compound form within the earth's crust.  
**Alkaline Earth Metals**
14. Have a strong unpleasant odor and are poisonous. **Halogens**
15. Many of these elements are man-made. **Rare Earth Elements**
16. Are the most non-reactive elements. **Noble Gases**
17. Elements in groups 3 through 12. **Transition Metals**

# Periodic Tables Most Wanted Worksheet Answers

**S Nieto**



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