

# PERIODIC TABLE

## Atomic Properties of the Elements

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Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data		1	
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd			
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of <sup>133</sup> Cs																							
speed of light in vacuum Planck constant elementary charge electron mass proton mass fine-structure constant Rydberg constant Boltzmann constant																							
$c$ $h$ $e$ $m_e$ $m_p$ $\alpha$ $R_\infty$ $k$																							
$299\,792\,458\text{ m s}^{-1}$ $6.626\,070\,15 \times 10^{-34}\text{ J s}$ $1.602\,176\,634 \times 10^{-19}\text{ C}$ $9.109\,383\,56 \times 10^{-31}\text{ kg}$ $1.672\,621\,92 \times 10^{-27}\text{ kg}$ $1/137.035\,999\,074$ $10\,973\,731.568\text{ m}^{-1}$ $1.380\,658\,4 \times 10^{-23}\text{ J K}^{-1}$																							
$1.380\,658\,4 \times 10^{-23}\text{ J K}^{-1}$																							

Atomic Number: 58  
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**  
Name: Cerium  
Atomic Weight: 140.116  
Ground-state Configuration: [Xe]4f<sup>1</sup>5d<sup>1</sup>6s<sup>2</sup>

\*Based upon <sup>12</sup>C. ( ) indicates the mass number of the longest-lived isotope.

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NIST SP 966 (September 2010)

# Periodix Table Study Packet

**Barron's Educational Series, Brian  
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