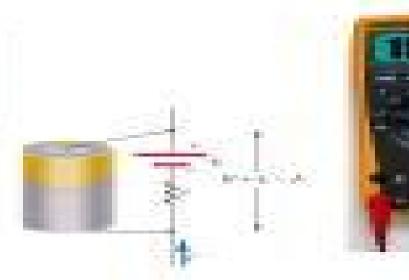
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manual to provide practical knowledge to the students related to various experiments in Physical Pharmaceutics I The manual mainly includes the experiments through which the students will learn how the basic properties of drug like solubility dissolution can limit the bioavailability of drug and in turn the efficacy of the drug The solubility enhancement techniques like reduction of particle size use of solubilizing agents or prodrugs etc are also discussed The students will also learn about the importance of drug stability and how the parameters like temperature pH humidity and light can affect the drug stability Physical pharmaceutics is very important subject which helps the researchers to design and formulate various dosage forms including its manufacturing aspects The students will be proficient in handling various equipment used in physical pharmaceutics laboratory The students will also be able to determine pKa partition coefficient surface tension critical micelle concentration HLB value stability constant buffer capacity etc in the laboratory. We have taken extra efforts to make the experiments simpler easy to understand and perform The graphs have also been provided wherever necessary Each experiment is divided into sections like aim practical significance relevant course outcomes practical skills relevant affective domain related outcomes practical outcomes minimum theoretical background requirements related questions and references The manual has been designed with more emphasis on the practical skill improvement of the students so that the students can perform the practical with ease and comfort We are very much thankful to the designer publisher printers and all the stakeholders for putting their efforts for successfully bringing this manual out for the students Hope this manual will help the students to learn the concept principles and perform the experiments We wish them all the best Physics Lab Manual Sesha Sai Kumar Vemula, 2017-07-23 Manual of Plane Geometry, on the Heuristic Plan George Irving Hopkins, 1891 Physical Laboratory Manual Emory Leon Chaffee, Frederick Albert Saunders, 1920 An Index to Undergraduate Science National Science Foundation (U.S.). Office of Experimental Projects and Programs, 1974

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