

Date: \_\_\_\_\_

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## Fission and Fusion

11. The number and mass of nuclei changes in a fission process. The number \_\_\_\_\_, whereas the mass \_\_\_\_\_.
12. How is the energy released? How about is the difference between the mass of the separate particles before and the products in a fission reaction?
13. Calculate the energy released when a nucleus of an isotope of Uranium splits into two smaller nuclei. Assume the mass of the fission product nuclei and the mass of the original nucleus are given.
14. Calculate the energy released when the nuclei of hydrogen and helium are fused to form a nucleus of Helium. Assume the mass of the hydrogen nuclei and the mass of the Helium nucleus are given.

$$E = mc^2 = \frac{9 \times 10^9 \text{ J kg}^{-1} \text{ m}^2 \text{ s}^{-2}}{(3 \times 10^8 \text{ m s}^{-1})^2} \times \Delta m \text{ kg}$$

$$E = 1 \text{ kg} \times (3 \times 10^8 \text{ m s}^{-1})^2 = 9 \times 10^{16} \text{ J} \quad \text{or} \quad 900 \text{ exajoules (EJ)}$$

15. Balance and/or the following nuclear reactions. Calculate the energy released or absorbed.

\_\_\_\_\_ Mass \_\_\_\_\_  $\rightarrow$   $^{235}_{92}\text{U} \rightarrow$   $^{141}_{54}\text{Xe} +$   $^{92}_{38}\text{Sr}$

\_\_\_\_\_ Mass \_\_\_\_\_  $\rightarrow$   $^{235}_{92}\text{U} +$   $^1_0\text{n} \rightarrow$   $^{141}_{54}\text{Xe}$

\_\_\_\_\_ Mass \_\_\_\_\_  $\rightarrow$   $^{235}_{92}\text{U} \rightarrow$   $^{141}_{54}\text{Xe} +$   $^{92}_{38}\text{Sr} +$   $2^1_0\text{n}$

\_\_\_\_\_ Mass \_\_\_\_\_  $\rightarrow$   $^{235}_{92}\text{U} \rightarrow$   $^{141}_{54}\text{Xe} +$   $^{92}_{38}\text{Sr}$

\_\_\_\_\_ Mass \_\_\_\_\_  $\rightarrow$   $^{235}_{92}\text{U} +$   $^1_0\text{n} \rightarrow$   $^{141}_{54}\text{Xe} +$   $^{92}_{38}\text{Sr} +$   $2^1_0\text{n}$  (How many neutrons are released?)

16. a. Complete the following is an example of a fusion reaction?

i. The mass of the \_\_\_\_\_

ii. The mass of the products \_\_\_\_\_

iii. Energy is released \_\_\_\_\_

iv. Energy is absorbed \_\_\_\_\_

17. a. The mass of the nuclei before is a given value. Assume the mass of the \_\_\_\_\_

i. The mass of the products \_\_\_\_\_

ii. Energy is released \_\_\_\_\_

iii. The mass of the products is the same as the mass of the \_\_\_\_\_

iv. Energy is released \_\_\_\_\_

18. Complete the following is an example of a fusion reaction?

Reaction:  $2^1_1\text{H} + 2^1_1\text{H} \rightarrow$   $^4_2\text{He} +$   $2^1_0\text{n}$  (How much energy is released?)

# Phys Sci Ch 14 Fission And Fusion Worksheet Answers

**Fiona Young-Brown**



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