

3RD TERM

Week: One

PARTICULATE NATURE OF MATTER

STRUCTURE OF MATTER

(a) EVIDENCE OF THE PARTICLE NATURE OF MATTER

The idea that matter is made up of minute particles called atom dated back to the ancient Greek. Atom is the smallest indivisible Particle of an element which can take part in a chemical change. It consists of proton, neutron and electron. Proton is positively charged, neutron is neutral and electron is negatively charged.

MOLECULES

A molecule is the smallest particle of a substance which can have a separate existence and still retains the properties of that substance.

BROWNIAN MOTION

Brownian motion is the rapid, constant and irregular motion of tiny particles. This can be observed when a dust particle is in continuous state of haphazard movement caused by the impact of moving water molecules. This continuous motion also occurs in smoke molecule.

DIFUSSION

Diffusion is the tendency of molecule to migrate and fill an empty space due to their random motions.

WEEK: Two

STATES OF MATTER

Matter can exist in three states- Solids, Liquids and Gases. The State of matter depends on Temperature and Pressure.

1. **SOLIDS:** The molecules of solids are closely packed together. They have definite shape and volume. Their molecules are arranged in a regular pattern and held rigidly by a strong intermolecular forces
2. **LIQUIDS:** Its molecule are freer than that of solid, they have definite volume but no definite shape. They have larger intermolecular distance than solid molecules.
3. **GASES:** The cohesive force binding the gases molecules together is negligible. They have no definite shape and Volume. Their molecules are at far distance apart. They have greater speed due to the space between its molecules. Their motion is linear except when they collide.

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Wolfgang Guggemos

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