

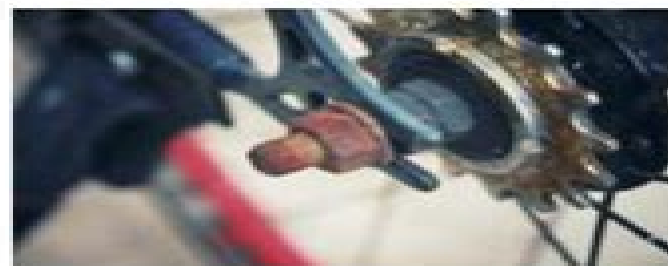
Physical and Chemical Changes

Now that you know the difference between a physical and chemical property, this information can be used to determine whether a change in matter can be identified as a physical or chemical change. Just like the properties of matter, physical and chemical changes of matter have different characteristics which can usually be easily identified.

<u>Physical Changes</u>	<u>Chemical Changes</u>
<ul style="list-style-type: none">• A change in size, shape, color, or a phase change (solid, liquid, gas)• No new substance is formed and the molecules do not change	<ul style="list-style-type: none">• A change in the physical and chemical properties• A new substance is formed and the molecules of the substance are changed
<u>Examples of Physical Changes</u>	<u>Examples of Chemical Changes</u>
<ol style="list-style-type: none">1. Melting a cube of sugar in coffee because it is still sugar4. Water changed to an ice cube, boiling water, or a melting ice cube, molecules remain the same5. Cutting apart an apple into smaller pieces making it easier to eat	<ol style="list-style-type: none">1. Burning a cube of sugar with fire changes its molecules4. Baking the ingredients of a cake changes many of the molecules due to the heat5. An apple beginning to rot or being digested inside a person's stomach after it is eaten, will both change its molecules

Chemical and physical changes take place every day all around you. If you change the way something looks, but have not made a new substance, it is a **physical change**. If the substance has been changed into another substance it is a **chemical change**.

In a chemical change the atoms making up the molecules change. There is a **chemical reaction** that takes place during a chemical change. The reaction takes place usually due to another substance or heat added to it. One of the most commonly seen chemical reaction is rust forming on metal. If you leave your bike in the rain, it may begin to rust. Iron is exposed to oxygen gas in the air which results in rust. The iron is **oxidized** (oxygen added) and becomes iron oxide, which is the chemical name for rust.



There are many, many other examples of physical and chemical changes. In summary, if the molecules stay the same it is a physical change, if the molecules change, it is a chemical change.



Name three of the same physical and chemical changes that take place in your life each day.

Learn
BRIGHT

Physical And Chemical Changes Lessons

M Lipman



Physical And Chemical Changes Lessons:

Unveiling the Magic of Words: A Report on "**Physical And Chemical Changes Lessons**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Physical And Chemical Changes Lessons**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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