Chapter 4 Atomic Structure

Section 4.2 The Structure of an Atom (pages 108–112)

This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.

Reading Strategy (page 108)

Monitoring Your Understanding Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

What I Know	What I Would	What I Have
About Atoms	Like to Learn	Learned
Most students will know that atoms are the "building blocks" of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

Properties of Subatomic Particles (pages 108-109)

1. What are three subatomic particles?

a. Protons b. Electrons c. Neutrons

Circle the letter that identifies a subatomic particle with a positive charge.

a. nucleus

(b) proton

c. neutron

d. electron

Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.

Comparing Subatomic Particles (pages 109-110)

Circle the letters of properties that vary among subatomic particles.

a. color

(b) mass

c. charge

d location in the atom

Circle the letter of the expression that accurately compares the masses of neutrons and protons.

a mass of 1 neutron = mass of 1 proton

b. mass of 2000 neutrons = mass of 1 proton

c. mass of 1 electron = mass of 1 proton

d. mass of 1 neutron = mass of 1 electron

Section 4 Studying Atoms Answers

RJ Shavelson

Section 4 Studying A	Atoms Answers:
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Adopting the Tune of Phrase: An Emotional Symphony within Section 4 Studying Atoms Answers

In a global used by displays and the ceaseless chatter of fast transmission, the melodic elegance and psychological symphony produced by the prepared term often fade into the back ground, eclipsed by the persistent noise and interruptions that permeate our lives. But, nestled within the pages of **Section 4 Studying Atoms Answers** a charming literary prize full of natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this interesting masterpiece conducts viewers on an emotional trip, skillfully unraveling the concealed melodies and profound impact resonating within each carefully crafted phrase. Within the depths of the touching examination, we can investigate the book is central harmonies, analyze their enthralling publishing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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