

# Stoichiometry

Quiz 11\*9 Section 9

## Section 9: Introduction to Stoichiometry

In the space provided, write the letter of the best response that best completes each problem or best answers each question.

1. Which of the following is the formula for a compound that does not, according to convention, use Greek prefixes when naming it?
  - a.  $\text{Sulfuric acid}$
  - b.  $\text{Carbon monoxide}$
  - c.  $\text{Sulfur pentachloride}$
  - d.  $\text{Phosphorus pentachloride}$
2. How many moles of oxygen are formed in the decomposition of 1 mole of  $\text{KClO}_3$ ?
  - a. 1 mole of  $\text{O}_2$  is formed.
  - b. 1.5 moles of  $\text{O}_2$  are formed.
  - c. 3 moles of  $\text{O}_2$  are formed.
  - d. 4.5 moles of  $\text{O}_2$  are formed.
3. The equation of a straight line is given as  $y = 2x + 3$ . Which of the following is the equation of the line perpendicular to this line?
  - a.  $y = 2x + 3$
  - b.  $y = -2x + 3$
  - c.  $y = -\frac{1}{2}x + 3$
  - d.  $y = \frac{1}{2}x + 3$
4. The chemical equation  $\text{C}_2\text{H}_6 + 7\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$  represents the combustion of ethane. Which of the following is the correct stoichiometric ratio for this reaction?
  - a. 1 mole of  $\text{C}_2\text{H}_6$  reacts with 7 moles of  $\text{O}_2$  to produce 2 moles of  $\text{CO}_2$  and 3 moles of  $\text{H}_2\text{O}$ .
  - b. 1 mole of  $\text{C}_2\text{H}_6$  reacts with 7 moles of  $\text{O}_2$  to produce 2 moles of  $\text{CO}_2$  and 3 moles of  $\text{H}_2\text{O}$ .
  - c. 1 mole of  $\text{C}_2\text{H}_6$  reacts with 7 moles of  $\text{O}_2$  to produce 2 moles of  $\text{CO}_2$  and 3 moles of  $\text{H}_2\text{O}$ .
  - d. 1 mole of  $\text{C}_2\text{H}_6$  reacts with 7 moles of  $\text{O}_2$  to produce 2 moles of  $\text{CO}_2$  and 3 moles of  $\text{H}_2\text{O}$ .
5. How many moles of  $\text{H}_2\text{O}$  are produced from the following chemical equation?
  - a. 1
  - b. 2
  - c. 3
  - d. 4



# Section Quiz Introduction To Stoichiometry

**Stephen Thompson**



## Section Quiz Introduction To Stoichiometry:

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This book delves into Section Quiz Introduction To Stoichiometry. Section Quiz Introduction To Stoichiometry is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Section Quiz Introduction To Stoichiometry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Section Quiz Introduction To Stoichiometry
- Chapter 2: Essential Elements of Section Quiz Introduction To Stoichiometry
- Chapter 3: Section Quiz Introduction To Stoichiometry in Everyday Life
- Chapter 4: Section Quiz Introduction To Stoichiometry in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Section Quiz Introduction To Stoichiometry. The first chapter will explore what Section Quiz Introduction To Stoichiometry is, why Section Quiz Introduction To Stoichiometry is vital, and how to effectively learn about Section Quiz Introduction To Stoichiometry.
3. In chapter 2, the author will delve into the foundational concepts of Section Quiz Introduction To Stoichiometry. This chapter will elucidate the essential principles that must be understood to grasp Section Quiz Introduction To Stoichiometry in its entirety.
4. In chapter 3, this book will examine the practical applications of Section Quiz Introduction To Stoichiometry in daily life. The third chapter will showcase real-world examples of how Section Quiz Introduction To Stoichiometry can be effectively utilized in everyday scenarios.
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6. In chapter 5, the author will draw a conclusion about Section Quiz Introduction To Stoichiometry. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Section Quiz Introduction To Stoichiometry.

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