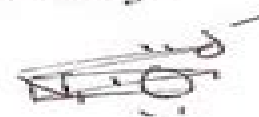
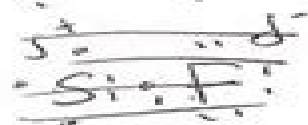
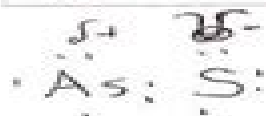


Electronegativity Worksheet

Using the table of electronegativities decide which type of bond should be formed between the two atoms.

Arsenic As 2.2	Sulfur S 2.5	Covalent / Ionic 1. $\Delta EN = 0.3$ Covalent
Cobalt Co 1.9	Bromine Br 2.8	2. 0.9 Polar Covalent
Germanium Ge 2.0	Selenium Se 2.6	3. 0.6 polar covalent
Silicon Si 1.9	Fluorine F 4.0	4. 2.1 ionic
Potassium K 0.8	Nitrogen N 3.0	5. 2.2 ionic
Nickel Ni 1.9	Oxygen O 3.5	6. 1.6 polar covalent
Barium Ba 0.9	Tin Sn 2.0	7. 1.1 polar covalent
Hydrogen H 2.1	Oxygen O 3.5	8. 1.4 polar covalent
Calcium Ca 1.3	Sulfur S 2.5	9. 1.2 polar covalent
Iron Fe 1.8	Carbon C 2.5	10. 0.7 polar covalent



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Karin Nielsen-Saines



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