

Periodic Trends and the Properties of the Elements Chemistry Lab #7

Name _____
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The periodic table is the most recognized symbol of chemistry across the world. It is a valuable tool that allows scientists not only to classify the elements but also to explain and predict their properties. Similarities and differences among the elements give rise to so-called periodic trends, both across rows and within columns of the periodic table. Recognizing periodic trends in the physical and chemical properties of the elements is key to understanding the full value of the periodic table.

Purpose

The purpose of this experiment is to identify periodic trends in the activity and solubility of the alkaline earth metals. In Part A, the reactions of magnesium, calcium, and aluminum with water and acids will be compared in order to determine the trend in metal activity within a group (Mg vs Ca) and across a period (Mg vs Al) in the periodic table. In Part B, the solubility of magnesium, calcium, strontium, and barium compounds will be studied and used to identify an unknown alkaline earth metal.

Pre-Lab Questions:

- 1) Read the entire procedure and safety precautions. Do you think extra pieces of magnesium and calcium metal should be disposed of down the drain? Why or why not?
- 2) The ionization energy of an element is defined as the amount of energy required to remove an electron from an individual atom. The following table gives the ionization energy (in units of kilojoules per mole) for five metals, listed in alphabetical order. Locate each of these metals on the periodic table and arrange them in order of rows and columns as in the periodic table.
 - a) Describe the trend in the ionization energy of elements within a group.
 - b) Describe the trend in the ionization energy of elements across a period.

| Metal | Calcium | Magnesium | Potassium | Sodium | Strontium |
|-------------------|---------|-----------|-----------|--------|-----------|
| Ionization Energy | 590 | 738 | 419 | 496 | 549 |

Materials

Aluminum foil
Magnesium ribbon
Calcium turnings
Sodium sulfate, 1 M Na_2SO_4
Magnesium chloride, 0.1 M MgCl_2
Strontium chloride, 0.1 M SrCl_2
Barium chloride, 0.1 M BaCl_2
Unknown metal chloride
Matches

HCl, 0.5 M
Potassium iodate, 0.2 M KIO_3
Sodium carbonate, 1 M Na_2CO_3
Calcium chloride, 0.1 M CaCl_2
Thermometer
Forceps
Litmus paper, red
Pipets
Reaction plate

Safety:

Calcium and magnesium are reactive, flammable, and possible skin irritants. HCl is toxic and corrosive to the skin and eyes.

Trends In The Periodic Table Lab Report

RC Schank



Trends In The Periodic Table Lab Report:

Handbook of Formative Assessment Heidi Andrade, Gregory J. Cizek, 2010-04-02 Formative assessment has recently become a focus of renewed research as state and federal policy makers realize that summative assessments have reached a point of diminishing returns as a tool for increasing student achievement. Consequently, supporters of large scale testing programs are now beginning to consider the potential of formative assessments to improve student achievement. The mission of this handbook is to comprehensively profile this burgeoning field of study. Written by leading international scholars and practitioners, each chapter includes a discussion of key issues that dominate formative assessment policy and practice today as well as those that are likely to affect research and practice in the coming years. Key features include: Comprehensive: nineteen chapters cover all aspects of formative assessment including classroom assessment, large scale applications, technological applications, applications for special needs students, K-12 and post secondary applications, psychometric considerations, case studies, and discussion of alternative assessment formats such as portfolios and performance assessments. Integrative: thoughtful attention is given to the integration of large scale and classroom assessments. Practical: provides practical guidance on how to conduct formative assessments that generate credible information to guide instruction. Global: provides perspectives from leading international scholars and practitioners whose expertise spans diverse settings, student populations, and educational systems. Accessible Style: although grounded in the latest research, the book's style and tone has been carefully crafted to make it accessible to both the textbook and professional markets. It will also be a critical reference book for researchers in teacher preparation, educational administration, and educational policy studies.

Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program. The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure, selecting the proper apparatus, employing the proper techniques systematically, analyzing and interpreting the data, and minimizing inherent variables. As a result of good data, a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data. Experiments write textbooks; textbooks don't write experiments. A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered. Further experimentation provides additional cause-effect observations leading to an even better understanding of the experiment. The 11th edition's experiments are informative and challenging while offering a solid foundation for technique, safety, and experimental procedure. The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment. The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit. An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment. Continued emphasis on handling

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