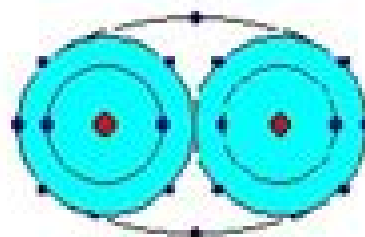
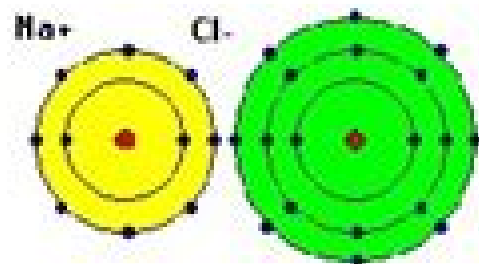


Section 8.2 – The Nature of Covalent Bonding

- ❑ In ionic bonding, atoms transfer electrons to achieve noble gas configuration.
- ❑ In covalent bonding, atoms share electrons to achieve noble gas configuration.
- ❑ Most atoms share electrons until they have a total of 8 valence electrons (octet rule). However, hydrogen only needs 2 electrons to be stable.



Section 8 2 Answers The Nature Of Covalent Bonding

Rachel Sandford



Section 8 2 Answers The Nature Of Covalent Bonding:

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