Chapter 9

Section 3 Limiting Reactants and Percentage Yield

Percentage Yield

- The theoretical yield is the maximum amount of product that can be produced from a given amount of reactant.
- The actual yield of a product is the measured amount of that product obtained from a reaction.
- The percentage yield is the ratio of the actual yield to the theoretical yield, multiplied by 100.

percentage yield = $\frac{\text{actual yield}}{\text{theorectical yield}} \times 100$



Chapter menu

Resources

Section Limiting Reactants And Percentage Yield Quiz

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

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