

## Balancing Equations Worksheet II

Count the number of atoms, then use coefficients to balance each equation.

- 1)  $\underline{\hspace{1cm}} \text{AlBr}_3 + \underline{\hspace{1cm}} \text{K} \rightarrow \underline{\hspace{1cm}} \text{KBr} + \underline{\hspace{1cm}} \text{Al}$
- 2)  $\underline{\hspace{1cm}} \text{FeO} + \underline{\hspace{1cm}} \text{PdF}_2 \rightarrow \underline{\hspace{1cm}} \text{FeF}_2 + \underline{\hspace{1cm}} \text{PdO}$
- 3)  $\underline{\hspace{1cm}} \text{P}_4 + \underline{\hspace{1cm}} \text{Br}_2 \rightarrow \underline{\hspace{1cm}} \text{PBr}_3$
- 4)  $\underline{\hspace{1cm}} \text{LiCl} + \underline{\hspace{1cm}} \text{Br}_2 \rightarrow \underline{\hspace{1cm}} \text{LiBr} + \underline{\hspace{1cm}} \text{Cl}_2$
- 5)  $\underline{\hspace{1cm}} \text{PbBr}_2 + \underline{\hspace{1cm}} \text{HCl} \rightarrow \underline{\hspace{1cm}} \text{HBr} + \underline{\hspace{1cm}} \text{PbCl}_2$
- 6)  $\underline{\hspace{1cm}} \text{CoBr}_3 + \underline{\hspace{1cm}} \text{CaSO}_4 \rightarrow \underline{\hspace{1cm}} \text{CaBr}_2 + \underline{\hspace{1cm}} \text{Co}_2(\text{SO}_4)_3$
- 7)  $\underline{\hspace{1cm}} \text{Na}_3\text{P} + \underline{\hspace{1cm}} \text{CaF}_2 \rightarrow \underline{\hspace{1cm}} \text{NaF} + \underline{\hspace{1cm}} \text{Ca}_3\text{P}_2$
- 8)  $\underline{\hspace{1cm}} \text{Mn} + \underline{\hspace{1cm}} \text{HI} \rightarrow \underline{\hspace{1cm}} \text{H}_2 + \underline{\hspace{1cm}} \text{MnI}_3$
- 9)  $\underline{\hspace{1cm}} \text{Li}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{NaBr} \rightarrow \underline{\hspace{1cm}} \text{Na}_3\text{PO}_4 + \underline{\hspace{1cm}} \text{LiBr}$
- 10)  $\underline{\hspace{1cm}} \text{CaF}_2 + \underline{\hspace{1cm}} \text{Li}_2\text{SO}_4 \rightarrow \underline{\hspace{1cm}} \text{CaSO}_4 + \underline{\hspace{1cm}} \text{LiF}$
- 11)  $\underline{\hspace{1cm}} \text{HBr} + \underline{\hspace{1cm}} \text{Mg}(\text{OH})_2 \rightarrow \underline{\hspace{1cm}} \text{MgBr}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$
- 12)  $\underline{\hspace{1cm}} \text{LiNO}_3 + \underline{\hspace{1cm}} \text{CaBr}_2 \rightarrow \underline{\hspace{1cm}} \text{Ca}(\text{NO}_3)_2 + \underline{\hspace{1cm}} \text{LiBr}$
- 13)  $\underline{\hspace{1cm}} \text{AgNO}_3 + \underline{\hspace{1cm}} \text{Li} \rightarrow \underline{\hspace{1cm}} \text{LiNO}_3 + \underline{\hspace{1cm}} \text{Ag}$
- 14)  $\underline{\hspace{1cm}} \text{Si}(\text{OH})_4 + \underline{\hspace{1cm}} \text{NaBr} \rightarrow \underline{\hspace{1cm}} \text{SiBr}_4 + \underline{\hspace{1cm}} \text{NaOH}$
- 15)  $\underline{\hspace{1cm}} \text{NaCN} + \underline{\hspace{1cm}} \text{CuCO}_3 \rightarrow \underline{\hspace{1cm}} \text{Na}_2\text{CO}_3 + \underline{\hspace{1cm}} \text{Cu}(\text{CN})_2$

**Challenge:**  $\underline{\hspace{1cm}} \text{C}_2\text{H}_6 + \underline{\hspace{1cm}} \text{O}_2 \rightarrow \underline{\hspace{1cm}} \text{CO}_2 + \underline{\hspace{1cm}} \text{H}_2\text{O}$

# Section Reactions And Equations Worksheet Answers

**Yicheng Fang**



## **Section Reactions And Equations Worksheet Answers:**

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This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry. It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry. Some of the worksheets are prepared along with model answers. Some other worksheets are meant for self assessment and evaluation purposes. It is also observed that some of the topics are specific to the referred curriculum. Some other topics are varyingly incorporated in other streams of study. Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations.

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**About the Author** DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone

Association and Au Sable Institute of Environmental Studies      **Prentice Hall Science Explorer** Michael J. Padilla, 2002  
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two presents linear ordinary differential equations in section 1 to include homogeneous and nonhomogeneous ODEs solving systems of ODEs using the matrix exponential and Laplace transform method In section two of chapter two nonlinear ordinary differential equations are presented and include simultaneous series reactions solving nonlinear ODEs with Maple's dsolve command stop conditions differential algebraic equations and steady state solutions Chapter three addresses boundary value problems

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*Revise A2 Chemistry for AQA* Paddy Gannon, 2005-04 Part of the series of AS and A2 revision guides this title gives students what they need to know for the AQA exams It includes material organised into bite sized chunks of information

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