8 • Electron Configurations & Periodicity

WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
	, <u>S</u>				S S S S S S S S S S S S S S S S S S S
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¹	¹⊠	180	180	'₩	'₩
Element: Ar # of e 's: 16	Element: Mg # of e-'s: 12	Element: N # of e-'s: 7	Element: Li # of e-'s: _3	Element: P # of e 's: 15	Element: Cl # of e ⁻ 's: 17

Write the electron configurations of each of these in long form and short form:

- Ar 152 252 206 352 306
 - AT TNET 352 3pt
- Mg 152 252 206 352
 - Mg [Ne] 352
- N 152 252 2p3 N [He] 252 2p3
- Li 152 25'
- P 152 252 Zpc 352 3p3
 - P [Ne] 35+ 3p3
- CI 152 252 2p6 352 3p5
- CI [Ne] 352 305



Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.

Fe 1s2 252 2p 352 3p 3d 452

Fe [Ar] 3d 452



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Table of Contents Section 5 3 Electron Configurations Answer Key

- 1. Understanding the eBook Section 5 3 Electron Configurations Answer Key
 - The Rise of Digital Reading Section 5 3 Electron Configurations Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Section 5 3 Electron Configurations Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 5 3 Electron Configurations Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Section 5 3 Electron Configurations Answer Key
 - Personalized Recommendations
 - Section 5 3 Electron Configurations Answer Key User Reviews and Ratings
 - Section 5 3 Electron Configurations Answer Key and Bestseller Lists

- 5. Accessing Section 5 3 Electron Configurations Answer Key Free and Paid eBooks
 - Section 5 3 Electron Configurations Answer Key Public Domain eBooks
 - Section 5 3 Electron Configurations Answer Key eBook Subscription Services
 - Section 5 3 Electron Configurations Answer Key Budget-Friendly Options
- 6. Navigating Section 5 3 Electron Configurations Answer Key eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Section 5 3 Electron Configurations Answer Key Compatibility with Devices
 - Section 5 3 Electron Configurations Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 5 3 Electron Configurations Answer Key
 - Highlighting and Note-Taking Section 5 3 Electron Configurations Answer Key
 - Interactive Elements Section 5 3 Electron Configurations Answer Key
- 8. Staying Engaged with Section 5 3 Electron Configurations Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 5 3 Electron Configurations Answer Key
- 9. Balancing eBooks and Physical Books Section 5 3 Electron Configurations Answer Key
 - ∘ Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Section 5 3 Electron Configurations Answer Key
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 5 3 Electron Configurations Answer Key
 - Setting Reading Goals Section 5 3 Electron Configurations Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 5 3 Electron Configurations Answer Key
 - Fact-Checking eBook Content of Section 5 3 Electron Configurations Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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