

REVIEW: Chapter 8 - Covalent Bonding

- Typically, how many bonds & lone pairs will each have when covalently bonding?
 - Boron – 3 bonds; 0 lone pairs
 - Silicone – 4 bonds; 0 lone pairs
 - Neon – 0 bonds; 4 lone pairs
 - Iodine – 1 bond; 3 lone pairs
 - Nitrogen – 3 bonds; 1 lone pair
 - Oxygen – 2 bonds; 2 lone pairs
- Which elements do not follow the octet rule? (Hint – there are four we went over)
 - ___H___ & ___B___ never follow the octet rule.
 - ___P___ & ___S___ sometimes don't follow the octet rule.
- What type of bond will each combination of atoms form?
 - F & N – 2 nonmetals, therefore covalent
 - Al & O – 1 metal/1 nonm., therefore ionic
 - Na & S – 1 metal/1 nonm., therefore ionic
 - B & Cl – 2 nonmetals, therefore covalent
 - Ca & I – 1 metal/1 nonm., therefore ionic
 - Fe & Cu – 2 metals, therefore metallic
- What is the difference between:
 - Molecules & Compounds?
Compounds can be made using all types of bonds (ionic, covalent or metallic), but molecules can only be made using covalent bonds
 - Monatomic elements & Diatomic molecules ?
Monatomic elements exist as single atoms; diatomic molecules are made up of two of the same atom
 - Structural formulas, Empirical formulas, & Molecular formulas?
All three are for covalent bonds only! Molecular show the actual number of atoms in the molecule; Empirical give the lowest whole number ratio; Structural formulas are not really formulas, they are drawing of the molecule.
 - Polar bonds & Non-polar bonds?
Both describe covalent bonds! In a polar bond the electrons are NOT shared equally; in a non-polar bond electrons are shared equally
- Classify each of the following as an atom or a molecule:
 - Be, atom
 - CO, molec.
 - N₂, molec.
 - H₂O, molec.
 - Ne, atom

Section 8 Covalent Bonding Workbook Answers

LP Steffe

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'LP Steffe' text.

Section 8 Covalent Bonding Workbook Answers:

This book delves into Section 8 Covalent Bonding Workbook Answers. Section 8 Covalent Bonding Workbook Answers is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Section 8 Covalent Bonding Workbook Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Section 8 Covalent Bonding Workbook Answers
 - Chapter 2: Essential Elements of Section 8 Covalent Bonding Workbook Answers
 - Chapter 3: Section 8 Covalent Bonding Workbook Answers in Everyday Life
 - Chapter 4: Section 8 Covalent Bonding Workbook Answers in Specific Contexts
 - Chapter 5: Conclusion
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 5. In chapter 4, this book will scrutinize the relevance of Section 8 Covalent Bonding Workbook Answers in specific contexts. This chapter will explore how Section 8 Covalent Bonding Workbook Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Section 8 Covalent Bonding Workbook Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Section 8 Covalent Bonding Workbook Answers.

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