

Nuclear Chemistry Worksheet

Identify the following types of nuclear radiation

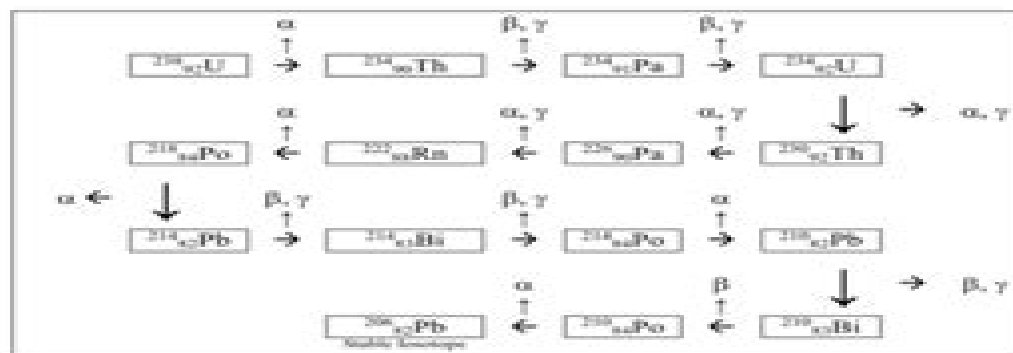
1. ${}_{-1}^0\text{e}$ _____
2. ${}_{2}^4\text{He}$ _____
3. ${}_{0}^0\gamma$ _____
4. Has no mass or charge, almost always follows α or β decay _____
5. Occurs when a neutron breaks apart and forms a proton _____
6. Least penetrating nuclear decay _____
7. Most damaging nuclear decay to the human body _____
8. Nuclear decay that can be stopped by skin or paper. _____
9. Nuclear decay that can be stopped by aluminum. _____
10. As a result, a proton is transformed into a neutron _____
11. The opposite of beta decay _____
12. Type of decay isotopes $>Z83$ will undergo _____

Complete the following nuclear equations.



Nuclear Decay Series

The figure maps the radioactive decay of uranium-238 to lead-206. Use the figure to answer the following questions.



19. How many alpha particles are produced as one atom of uranium-238 decays to an atom of lead-206?

20. How many beta particles? _____

21. Explain why lead-206 is a stable isotope. _____

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