NUCLEAR DECAY

Name:_____

Predict the products of the following nuclear reactions.

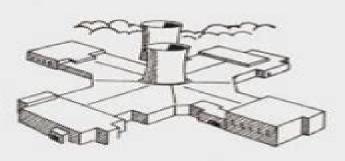
3.
$$\frac{235}{92}U \rightarrow \frac{4}{2}He + \frac{231}{90}Th$$

9.
$$\frac{235}{92}$$
 + $\frac{1}{9}$ n $\rightarrow \frac{140}{55}$ Ba + $\frac{91}{55}$ Kr + $\frac{31}{9}$ n

12. Plutonium-239 can be produced by bombarding uranium-238 with alpha particles. How many neutrons will be produced as a by product of each reaction? Write the nuclear equation for this reaction.

 Neutron bombardment of plutonium-239 yields americium-240 and another particle. Write the nuclear equation and identify the other particle produced.

14. Alpha-particle bomburdment of plutonium-239 produces a neutron and another isotope. Write the nuclear equation for this reaction and identify the isotope.



Nuclear Decay Equations Answers

Barron's Educational Series, Albert Tarendash

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