

## BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction ( $\alpha$  emission,  $\beta$  emission,  $\gamma$  emission, positron emission, artificial transmutation, fission, or fusion) described. (30 points)

| Write the missing product or reactant   | Type of Nuclear Reaction |
|---|--------------------------|
| 1. ${}_{19}^{42}\text{K} \rightarrow {}_{-1}^0\text{e} + \underline{\hspace{2cm}}$  | <hr/>                    |
| 2. ${}_{94}^{239}\text{Pu} \rightarrow {}_2^4\text{He} + \underline{\hspace{2cm}}$  | <hr/>                    |
| 3. ${}_{92}^{235}\text{U} \rightarrow \underline{\hspace{2cm}} + {}_{90}^{231}\text{Th}$                                    | <hr/>                    |
| 4. ${}_1^1\text{H} + {}_1^3\text{H} \rightarrow \underline{\hspace{2cm}}$   | <hr/>                    |
| 5. ${}_3^6\text{Li} \rightarrow {}_0^1\text{n} + {}_2^4\text{He} + \underline{\hspace{2cm}}$                                | <hr/>                    |
| 6. ${}_{13}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{15}^{30}\text{P} + \underline{\hspace{2cm}}$                    | <hr/>                    |
| 7. ${}_4^9\text{Be} + {}_1^1\text{H} \rightarrow \underline{\hspace{2cm}} + {}_2^4\text{He}$                                | <hr/>                    |
| 8. ${}_{19}^{37}\text{K} \rightarrow {}_{+1}^0\text{e} + \underline{\hspace{2cm}}$  | <hr/>                    |
| 9. $\underline{\hspace{2cm}} + {}_0^1\text{n} \rightarrow {}_{56}^{142}\text{Ba} + {}_{36}^{91}\text{Kr} + 3{}_0^1\text{n}$ | <hr/>                    |
| 10. ${}_{92}^{238}\text{U} + {}_2^4\text{He} \rightarrow \underline{\hspace{2cm}} + {}_0^1\text{n}$                         | <hr/>                    |
| 11. ${}_6^{14}\text{C} \rightarrow {}_7^{14}\text{N} + \underline{\hspace{2cm}}$  | <hr/>                    |
| 12. ${}_{75}^{187}\text{Re} + \underline{\hspace{2cm}} \rightarrow {}_{75}^{188}\text{Re} + {}_1^1\text{H}$                 | <hr/>                    |
| 13. ${}_{11}^{22}\text{Na} + \underline{\hspace{2cm}} \rightarrow {}_{10}^{22}\text{Ne}$                                    | <hr/>                    |
| 14. ${}_{84}^{218}\text{Po} \rightarrow \underline{\hspace{2cm}} + {}_2^4\text{He}$   | <hr/>                    |
| 15. ${}_{99}^{253}\text{Es} + {}_2^4\text{He} \rightarrow {}_0^1\text{n} + \underline{\hspace{2cm}}$                        | <hr/>                    |

# Nuclear Reactions Study Guide Answer Sheet

**Donald N. Niederkorn**



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