Chapter 19 - Nuclear Chemistry

This packet contains background and explanatory information on Nuclear Chemistry in addition to practice handouts. The information should be read and used to complete all of the practice included. I will be collecting this packet and giving points for the completion of the contents. You may want to consult your textbook (Chapter 19) or online resources for addition explanations and examples if needed.

Chapter Objectives: The student will (be able to)...

- Compare and contrast alpha, beta, and gamma radiation.
- Identify alpha decay, beta decay, fission, and fusion reactions.
- Balance nuclear equations.
- · Solve half-life calculations.

Part I: Radioactivity - What is it? http://www.darvill.clara.net/mucrad/index.htm

All substance are made of atoms. These have electrons (e) around the outside, and a nucleus in the middle.

The nucleus consists of protons (p) and neutrons (n), and is extremely small. (Atoms are almost entirely made of empty space!)

In some types of atom, the nucleus is unstable, and will decay into a more stable atom. This radioactive decay is completely spontancous. The energy that is released from the nucleus of the atom is radiation.

When an unstable nucleus decays, there are three ways that it can do so. It may give out:-

- an alpha particle (we use the symbol α)
- a beta particle (symbol β)
- a gamma ray (symbol γ)

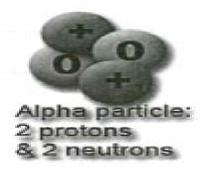
Many radioactive substances emit $\alpha_{\text{particles and}} \beta_{\text{particles as well as}} \gamma_{\text{rays}}$. In fact, you won't find a pure $\gamma_{\text{source: anything that gives off}} \gamma_{\text{rays will also give off}} \alpha_{\text{and/or}} \beta_{\text{too.}}$

A. Alpha Particles - More Information

- Alpha particles are made of 2 protons and 2 neutrons.
- This means that they have a charge of +2, and a mass of 4 (the mass is measured in "atomic mass units", where each proton & neutron=1)

We can write them as ${}^{4}_{2}$ Ct, or, because they're the same as a helium nucleus.

- Alpha particles are relatively slow and heavy.
- They have a low penetrating power you can stop them with just a sheet of paper.
- Because they have a large charge, alpha particles ionize (pull electrons from) other atoms strongly.



Nuclear Chemistry Packet Answer Key

S Ben Porath

Nuclear Chemistry Packet Answer Key:

Essavs in Contemporary Chemistry Gerhard Quinkert, 2001 In this volume internationally renowned chemists recount their roles in the progress of chemistry research toward elucidation of biological processes Beginning with a historical perspective on the development of X ray crystallography the reader is regaled with first hand accounts of research milestones Included are descriptions of the cutting edge nuclear magnetic resonance and electron paramagnetic resonance spectroscopic techniques the dynamics of ultra fast reactions and the central problem of molecular oxygen activation in biological processes The roles of chiral auxiliaries in organic synthesis and of photochemistry in natural product synthesis are told and innovations in combinatorial synthesis approaches are described Contemporary issues in molecular recognition and modulation of molecular function are addressed concluding with a missive regarding how the frontiers of medical ethics may be breached by molecular manipulations. The contributors who number among the finest scientists in the world including two Nobel Prize winners are Peter B Dervan Jack D Dunitz Christian Griesinger Jean Marie Lehn Thomas F Prisner Gerhard Quinkert Peter G Schultz Helmut Schwarz Dieter Seebach and Ahmed Zewail Additionally there is a prologue by Albert Eschenmoser for whom this collection was conceived and an epilogue that contains facsimiles of notes from his landmark lecture Synthesis of Co Enzyme B12 A Vehicle for Teaching Organic Synthesis This book is definitely a must for all who want to read or to read again where we stand in our chemical comprehension of the fascinating relationship between chemical structure and biological processes how we got here and what the future might hold Oswaal NCERT Textbook Solution Class 11 | Physics | Chemistry | Biology | Set of 3 Books | For Latest Exam Oswaal Editorial Board, 2024-03-30 Description of the Product Updated for 2024 25 The books are 100% updated for the academic year 2024 25 adhering strictly to the latest NCERT quidelines Comprehensive Coverage We cover all concepts and topics outlined in the most recent NCERT textbooks Visual Learning Aids Explore theoretical concepts and concept videos that offer a brief description of the topic and help visualize complex concepts Effective Revision Tools Benefit from crisp Revision Notes Mind Maps and Mnemonics designed to facilitate efficient and effective review Complete Question Coverage All questions from the NCERT textbooks are covered Resources in Education ,1994-04 in our solutions providing a thorough grasp of the subject matter **Ouantum** Chemistry and Dynamics of Excited States Leticia González, Roland Lindh, 2021-02-01 An introduction to the rapidly evolving methodology of electronic excited states For academic researchers postdocs graduate and undergraduate students Quantum Chemistry and Dynamics of Excited States Methods and Applications reports the most updated and accurate theoretical techniques to treat electronic excited states From methods to deal with stationary calculations through time dependent simulations of molecular systems this book serves as a guide for beginners in the field and knowledge seekers alike Taking into account the most recent theory developments and representative applications it also covers the often overlooked gap between theoretical and computational chemistry An excellent reference for both researchers and students

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