

1. Base your answer to the following question on Given the nuclear equation:



The particle represented by  $X$  is

- A)  ${}^9_4\text{Li}$  B)  ${}^9_4\text{Be}$  C)  ${}^{10}_5\text{B}$  D)  ${}^{10}_6\text{C}$
2. Which isotope will spontaneously decay and emit particles with a charge of +2?
- A)  ${}^{59}\text{Fe}$  B)  ${}^{137}\text{Cs}$  C)  ${}^{198}\text{Au}$  D)  ${}^{220}\text{Fr}$
3. Which of these particles has the greatest mass?
- A) alpha B) beta  
C) neutron D) positron
4. Base your answer to the following question on Given the nuclear equation:



What particle is represented by  $X$ ?

- A) alpha B) beta  
C) neutron D) positron
5. Alpha particles and beta particles differ in
- A) mass, only  
B) charge, only  
C) both mass and charge  
D) neither mass nor charge
6. Which statement best describes gamma radiation?
- A) It has a mass of 1 and a charge of 1.  
B) It has a mass of 0 and a charge of -1.  
C) It has a mass of 0 and a charge of 0.  
D) It has a mass of 4 and a charge of +2.
7. Which type of radiation is most similar to high-energy x-rays?
- A) alpha B) beta  
C) neutron D) gamma
8. What is the total number of years that must pass before only 25.00 grams of an original 100.0-gram sample of C-14 remains unchanged?
- A) 2865 y B) 5730 y  
C) 11 460 y D) 17 190 y
9. Which isotope is most commonly used in the radioactive dating of the remains of organic materials?
- A)  ${}^{14}\text{C}$  B)  ${}^{16}\text{N}$  C)  ${}^{32}\text{P}$  D)  ${}^{37}\text{K}$

10. Based on Reference Table N, what fraction of a radioactive  ${}^{90}\text{Sr}$  sample would remain unchanged after 56.2 years?

A)  $\frac{1}{2}$  B)  $\frac{1}{4}$  C)  $\frac{1}{8}$  D)  $\frac{1}{16}$

11. What is the half-life and decay mode of Rn-222?

A) 1.91 days and alpha decay  
B) 1.91 days and beta decay  
C) 3.82 days and alpha decay  
D) 3.82 days and beta decay

12. After 32 days, 5 milligrams of an 80-milligram sample of a radioactive isotope remains unchanged. What is the half-life of this element?

A) 8 days B) 2 days  
C) 16 days D) 4 days

13. Which reaction converts an atom of one element to an atom of another element?

A) combustion B) polymerization  
C) saponification D) transmutation

14. Radioactive cobalt-60 is used in radiation therapy treatment. Cobalt-60 undergoes beta decay. This type of nuclear reaction is called

A) natural transmutation  
B) artificial transmutation  
C) nuclear fusion  
D) nuclear fission

15. In which reaction is mass converted to energy by the process of fission?

A)  ${}^{14}_7\text{N} + {}^1_0\text{n} \rightarrow {}^{14}_6\text{C} + {}^1_1\text{H}$   
B)  ${}^{235}_{92}\text{U} + {}^1_0\text{n} \rightarrow {}^{87}_{35}\text{Br} + {}^{146}_{57}\text{La} + 3{}^1_0\text{n}$   
C)  ${}^{226}_{88}\text{Ra} \rightarrow {}^{222}_{86}\text{Ra} + {}^4_2\text{He}$   
D)  ${}^2_1\text{H} + {}^2_1\text{H} \rightarrow {}^4_2\text{He}$

16. Which statement best describes what happens in a fission reaction?

A) Heavy nuclei split into lighter nuclei.  
B) Light nuclei form into heavier nuclei.  
C) Energy is released and less stable elements are formed.  
D) Energy is absorbed and more stable elements are formed.

# Nuclear Chemistry Review Packet

**Sabine Zange**



## **Nuclear Chemistry Review Packet:**

Femtochemistry Ahmed H. Zewail, 1994 These two volumes on Femtochemistry present a timely contribution to a field central to the understanding of the dynamics of the chemical bond This century has witnessed great strides in time and space resolutions down to the atomic scale providing chemists biologists and physicists with unprecedented opportunities for seeing microscopic structures and dynamics Femtochemistry is concerned with the time resolution of the most elementary motions of atoms during chemical change bond breaking and bond making on the femtosecond 10<sup>-15</sup> second time scale This atomic scale of time resolution has now reached the ultimate for the chemical bond and as Lord George Porter puts it chemists are near the end of the race against time These two volumes cover the general concepts techniques and applications of femtochemistry Professor Ahmed Zewail who has made the pioneering contributions in this field has from over 250 publications selected the articles for this anthology These volumes begin with a commentary and a historical chronology of the milestones He then presents a broad perspective of the current state of knowledge in femtochemistry by researchers around the world and discusses possible new directions In the words of a colleague it is a must on the reading list for all of my students all readers will find this to be an informative and valuable overview The introductory articles in Volume I provide reviews for both the non experts as well as for experts in the field This is followed by papers on the basic concepts For applications elementary reactions are studied first and then complex reactions Volume I is complete with studies of solvation dynamics non reactive systems ultrafast electron diffraction and the control of chemical reactions Volume II continues with reaction rates the concept of elementary intramolecular vibrational energy redistribution IVR and the phenomena of rotational coherence which has become a powerful tool for the determination of molecular structure via time resolution The second volume ends with an extensive list of references according to topics based on work by Professor Zewail and his group at Caltech These collected works by Professor Zewail will certainly be indispensable to both experts and beginners in the field The author is known for his clarity and for his creative and systematic contributions These volumes will be of interest and should prove useful to chemists biologists and physicists As noted by Professor J Manz Berlin and Professor A W Castleman Jr Penn State femtochemistry is yielding exciting new discoveries from analysis to control of chemical reactions with applications in many domains of chemistry and related fields e g physical organic and inorganic chemistry surface science molecular biology etc

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**Quantum Chemistry and Dynamics of Excited States** Leticia González,Roland Lindh,2020-11-10 An introduction to the rapidly evolving methodology of electronic excited states For academic researchers postdocs graduate and undergraduate students Quantum Chemistry and Dynamics of Excited States Methods and Applications reports the most updated and accurate theoretical techniques to treat electronic excited states From methods to deal with stationary calculations through time dependent simulations of molecular systems this book serves as a guide for beginners in the field and knowledge seekers alike Taking into account the most recent theory developments and representative applications it also covers the often overlooked gap between theoretical and computational chemistry An excellent reference for both researchers and students Excited States provides essential knowledge on quantum chemistry an in depth overview of the latest developments and theoretical techniques around the properties and nonadiabatic dynamics of chemical systems Readers will learn Essential theoretical techniques to describe the properties and dynamics of chemical systems Electronic Structure methods for stationary calculations Methods for electronic excited states from both a quantum chemical and time dependent point of view A breakdown of the most recent developments in the past 30 years For those searching for a better understanding of excited states as they relate to chemistry biochemistry industrial chemistry and beyond Quantum Chemistry and Dynamics of Excited States provides a solid education in the necessary foundations and important theories of excited states in photochemistry and ultrafast phenomena

**Advances in Atomic, Molecular, and Optical Physics** ,2023-06-01 Advances in Atomic Molecular and Optical Physics Volume 72 highlights new advances in the field with this new volume presenting interesting chapters written by an international board of authors Provides the authority and expertise of leading contributors from an international board of authors Presents the latest release in the Advances in Atomic Molecular and Optical Physics series Includes the latest information in the field

**Energy Research Abstracts** ,1993

*Desorption Induced by Electronic Transitions DIET V* Alan R. Burns, Ellen B. Stechel, Dwight R. Jennison, 2013-03-07 This volume in the Springer Series on Surface Sciences presents a recent account of advances in the ever broadening field of electron and photon stimulated surface processes As in previous volumes these advances are presented as the proceedings of the International Workshop on Desorption Induced by Electronic Transitions the fifth workshop DIET V was held in Taos New Mexico April 1 4 1992 It will be abundantly clear to the reader that DIET is not restricted to desorption but has for several years included photochemistry non thermal surface modification exciton self trapping and many other phenomena that are induced by electron or photon bombardment However most stimulated surface processes do share a common physics initial electronic excitation localization of the excitation and conversion of electronic energy into nuclear kinetic

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**Time-Resolved Spectroscopy** Thomas Weinacht, Brett J. Pearson, 2018-12-21 This concise and carefully developed text offers a reader friendly guide to the basics of time resolved spectroscopy with an emphasis on experimental implementation The authors carefully explain and relate for the reader how measurements are connected to the core physical principles They use the time dependent wave packet as a building block for understanding quantum dynamics progressively advancing to more complex topics The topics are discussed in paired sections one discussing the theory and the next presenting the related experimental methods A wide range of readers including students and newcomers to the field will gain a clear and practical understanding of how to measure aspects of molecular dynamics such as wave packet motion intramolecular vibrational relaxation and electron electron coupling and how to describe such measurements mathematically

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**Controlled Thermonuclear Research** United States. Congress. Joint Committee on Atomic Energy. Subcommittee on Research, Development, and Radiation, 1972      Science & Technology Review ,1997      *Annual Review* □□□□□□□□□□□□□□□□  
□,2012      **Laboratory Astrochemistry** Stephan Schlemmer, Thomas Giesen, Harald Mutschke, 2015-04-27 Written by leading scientists in the field and intended for a broader readership this is an ideal starting point for an overview of current research and developments As such the book covers a broad spectrum of laboratory astrophysics and chemistry describing recent advances in experiments as well as theoretical work including fundamental physics and modeling chemical networks For researchers as well as students and newcomers to the field      *Research Project Selection* United States. Congress. House. Committee on Science and Technology. Task Force on Science Policy, 1986      *Handbook of Computational Chemistry* Jerzy Leszczynski, 2012-01-13 The role the Handbook of Computational Chemistry is threefold It is primarily intended to be used as a guide that navigates the user through the plethora of computational methods currently in use it explains their limitations and advantages and it provides various examples of their important and varied applications This reference work is presented in three volumes Volume I introduces the different methods used in computational chemistry Basic assumptions common to the majority of computational methods based on molecular quantum or statistical mechanics are outlined and special attention is paid to the limits of their applicability Volume II portrays the applications of computational methods to model systems and discusses in detail molecular structures the modelling of various properties of molecules and chemical

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