

Nuclear Chemistry Webquest

Name _____

In this webquest, you will explore nuclear chemistry in real-world situations. You will learn about fusion and fission, types of radiation, its effects on humans, and how nuclear power is produced as well as its repercussions and disasters. Follow the steps below.

1. Enter the following web address: www.rcsdl2.org/Page/37614
2. Under web, click on Nuclear Chemistry to open a copy of this document
3. Click on the links provided to answer the questions in each section

Introduction to Atomic Physics

1. _____ is the smallest particle of matter that maintains the properties of that element.
2. Sir Ernst Rutherford concluded what two things about atomic structure:
 - a. _____
 - b. _____
3. Niels Bohr contributed what to atomic structure theory:
 - a. _____
4. What is the unit of measure for the diameter of atoms? _____
5. What is the unit of measure from above equal to? _____
6. An atom consists of three basic subatomic particles. List each below. Include its charge and who discovered it.
 - a. _____ ; _____ ; _____
 - b. _____ ; _____ ; _____
 - c. _____ ; _____ ; _____
7. The number of which subatomic particles determines the chemical element that it is? _____
8. The number and arrangement of which subatomic particle determines most of the atom's properties? _____
9. Protons and neutrons are composed of even smaller subatomic particles called _____.
10. Define isotopes: _____
11. Isotopes have the same _____ properties but very different _____ properties.
12. Most isotopes are stable but some are _____
13. What is binding energy? _____
14. What does binding energy determine? _____

15. Einstein's famous equation of relativity, $E = mc^2$, explains the relationship between the binding energy and mass defect. It explains that a small amount of _____ can produce a large amount of _____.
16. Define radioactivity: _____
17. Name the three scientists that discovered and researched radioactivity:
 - a. _____
 - b. _____
 - c. _____

Nuclear Chemistry Webquest Worksheet

Sally Ann Vonderbrink

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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