

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Nuclear Decay

**Vocabulary:** alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The chart below gives the locations, charges, and approximate masses of three **subatomic particles**. The approximate mass of each particle is given in universal mass units (u).

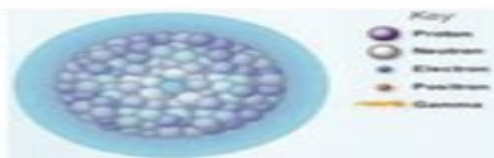
Particle	Location	Charge	Approximate mass
Proton	Nucleus	$1^+$	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	$1^-$	0 u

- The **mass number** of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom? \_\_\_\_\_
- The **atomic number** of an element is equal to the number of protons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium? \_\_\_\_\_

### Gizmo Warm-up

While most atoms are stable, some are **radioactive**, which means that they have a tendency to undergo spontaneous **nuclear decay**. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the Nuclear Decay Gizmo™. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



- Click **Play** (▶) and then click **Pause** (⏸) when the **alpha particle** is clearly visible. What is an alpha particle made of? \_\_\_\_\_
- Click **Play** and observe. Besides the alpha particle, what else is emitted from the nucleus during alpha decay? \_\_\_\_\_

**Gamma rays** are energetic electromagnetic waves; they are often emitted in nuclear decay.

### Activity A: Alpha decay

Get the Gizmo ready:

- Click **Reset** (↺).
- Check that **Alpha decay** and **Uranium** are selected.



**Question:** How does alpha decay change the nucleus of a radioactive atom?

- Predict:** As you observed in the warm-up activity, an alpha particle consists of two protons and two neutrons. How will the emission of an alpha particle affect the following?
  - The atomic number of the atom: \_\_\_\_\_
  - The mass number of the atom: \_\_\_\_\_
- Calculate:** Turn on **Write equation**. What you see is an equation that shows the original uranium atom on the left. The boxes on the right represent the **daughter product**—the atom produced by radioactive decay—and the emitted alpha particle.
  - In the top left box, write the mass number of the daughter product and press "Enter" on your keyboard. What is this number? \_\_\_\_\_

# Nuclear Decay Gizmo Worksheet Answers

**Donald Douglas Burgess, McMaster  
University. Department of Chemistry**



## **Nuclear Decay Gizmo Worksheet Answers:**

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**Handbook of Nuclear Decay Modes** Dorin N. Poenaru,1993-11-01      **Nuclear Decay Curve Analysis** Ditmar Bock,1969      **Evidence for Variation in Nuclear Decay Rates** ,2012      **Recommended Nuclear Decay Data** Andy Pearce,National Physical Laboratory (Great Britain). Division of Quality of Life,2008      *NUCLEAR REACTIONS* NARAYAN CHANGDER,2024-04-08 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject

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**Nuclear Decay Scheme Studies of  $^{109}\text{d}^{127}\text{mTe}$ ,  $^{9.4}\text{-h}^{127}\text{gTe}$  and  $^{2.2}\text{-h}^{127}\text{gSn}$**  Kenneth Ellis Apt,1971

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