



Nuclear Radiation

Section Review

Objectives

1. Calculate the activity of a radioactive substance.
2. Calculate the time taken for a given amount of a radioactive substance to decay.

Terminology

- | | | |
|-------------------|-------------------|-------------------|
| 1. radioactivity | 2. half-life | 3. activity |
| 4. disintegration | 5. decay constant | 6. parent nuclide |

Part A: Completion

Fill in the appropriate word or phrase in each of the following sentences. Use the words in the list to complete the sentences. Each word can be used only once.

Isotopes with an unstable nucleus are **1** and are called **2**. The **3** of a radioactive substance is the **4** of disintegrations per second of a **5**. The substance may be given **6** or given **7** without causing an alpha particle to penetrate **8**, meaning that it is easily stopped by a sheet of paper. Some substances undergo **9** from one position to another. **10** is represented by the Greek letter λ . The half-life of a substance is the time taken for **11** of the substance to decay. The half-life of a substance is **12** by **13**. The half-life of a substance is **14** by **15**. The half-life of a substance is **16** by **17**. The half-life of a substance is **18** by **19**. The half-life of a substance is **20** by **21**.

1. Radioactive
2. Radioisotopes
3. Half-life
4. Activity
5. Sample
6. Decay
7. Disintegration
8. Alpha
9. Beta
10. Gamma
11. Decay
12. Half-life
13. Decay
14. Half-life
15. Decay
16. Half-life
17. Decay
18. Half-life
19. Decay
20. Half-life

Part B: True-False

Change each of the following statements to either true (T) or false (F).

- | | |
|------------------------------------------------------------------------------|------------------------------------------------------------------------------|
| 1. The activity of a substance is the number of disintegrations per second. | 11. The activity of a substance is the number of disintegrations per second. |
| 2. The activity of a substance is the number of disintegrations per second. | 12. The activity of a substance is the number of disintegrations per second. |
| 3. The activity of a substance is the number of disintegrations per second. | 13. The activity of a substance is the number of disintegrations per second. |
| 4. The activity of a substance is the number of disintegrations per second. | 14. The activity of a substance is the number of disintegrations per second. |
| 5. The activity of a substance is the number of disintegrations per second. | 15. The activity of a substance is the number of disintegrations per second. |
| 6. The activity of a substance is the number of disintegrations per second. | 16. The activity of a substance is the number of disintegrations per second. |
| 7. The activity of a substance is the number of disintegrations per second. | 17. The activity of a substance is the number of disintegrations per second. |
| 8. The activity of a substance is the number of disintegrations per second. | 18. The activity of a substance is the number of disintegrations per second. |
| 9. The activity of a substance is the number of disintegrations per second. | 19. The activity of a substance is the number of disintegrations per second. |
| 10. The activity of a substance is the number of disintegrations per second. | 20. The activity of a substance is the number of disintegrations per second. |

Nuclear Radiation Section Review

H Kauffman



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