

NUCLEAR CHEMISTRY WORKSHEET

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Most nuclear reactions involve changes in the number of protons and/or neutrons in the nucleus of an atom. These changes are called transmutations because an atom of one element is changed into an atom of a different element. During these changes, nuclear particles may be absorbed and/or emitted along with a release of energy.

There are four types of nuclear reactions:

- 1) *natural radioactive decay*: transmutations occur by natural radioactivity (the ability of a nucleus to emit a nuclear particle and energy without external stimulation);
- 2) *artificial transmutation*: transmutations occur by nuclear disintegration caused by external stimulation as a scientist bombards a nucleus with a particle (the addition of another nuclear particle makes the nucleus unstable);
- 3) *fission*: certain nuclei having a large mass are bombarded with special particles that cause the nuclei to split into two nuclei each having a smaller mass;
- 4) *fusion*: nuclei of light elements are combined to form heavier nuclei.

Nuclear reactions are written using symbols in the A_ZX notation. The symbols for the major nuclear particles involved in nuclear reactions are given below.

Alpha particle	${}^4_2\text{He}$ (an alpha particle is a helium nucleus)
Proton	${}^1_1\text{H}$ (the most common hydrogen nucleus is a proton)
Neutron	${}^1_0\text{n}$
Electron	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ (also called a beta particle)
Positron	${}^0_1\text{e}$ or ${}^0_1\beta$
Gamma ray	${}^0_0\gamma$

The table below contains information about nuclear reactions involving the emission of radiation.

NUCLEAR CHANGES			
Type of Reaction	Radiation (Particle)	Effect on the Atomic Number	Effect on the Atomic Mass
Alpha emission (α)	${}^4_2\text{He}$	decrease by 2	decrease by 4
Beta emission (β)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	increase by 1	no change
Positron emission (β^+)	${}^0_1\text{e}$ or ${}^0_1\beta$	decrease by 1	no change
Gamma emission	${}^0_0\gamma$	no change	no change
Electron capture (EC)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	decrease by 1	no change

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Table of Contents Nuclear Power Worksheet With Answers Ch 22 1

1. Understanding the eBook Nuclear Power Worksheet With Answers Ch 22 1
 - The Rise of Digital Reading Nuclear Power Worksheet With Answers Ch 22 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Nuclear Power Worksheet With Answers Ch 22 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Power Worksheet With Answers Ch 22 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nuclear Power Worksheet With Answers Ch 22 1
 - Personalized Recommendations
 - Nuclear Power Worksheet With Answers Ch 22 1 User Reviews and Ratings
 - Nuclear Power Worksheet With Answers Ch 22 1 and Bestseller Lists

5. Accessing Nuclear Power Worksheet With Answers Ch 22 1 Free and Paid eBooks
 - Nuclear Power Worksheet With Answers Ch 22 1 Public Domain eBooks
 - Nuclear Power Worksheet With Answers Ch 22 1 eBook Subscription Services
 - Nuclear Power Worksheet With Answers Ch 22 1 Budget-Friendly Options
6. Navigating Nuclear Power Worksheet With Answers Ch 22 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Nuclear Power Worksheet With Answers Ch 22 1 Compatibility with Devices
 - Nuclear Power Worksheet With Answers Ch 22 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nuclear Power Worksheet With Answers Ch 22 1
 - Highlighting and Note-Taking Nuclear Power Worksheet With Answers Ch 22 1
 - Interactive Elements Nuclear Power Worksheet With Answers Ch 22 1
8. Staying Engaged with Nuclear Power Worksheet With Answers Ch 22 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Power Worksheet With Answers Ch 22 1
9. Balancing eBooks and Physical Books Nuclear Power Worksheet With Answers Ch 22 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Power Worksheet With Answers Ch 22 1
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Nuclear Power Worksheet With Answers Ch 22 1
 - Setting Reading Goals Nuclear Power Worksheet With Answers Ch 22 1
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nuclear Power Worksheet With Answers Ch 22 1
 - Fact-Checking eBook Content of Nuclear Power Worksheet With Answers Ch 22 1
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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