Nuclear Fission and Fusion

Why?

Fission and fusion are two processes that alter the nucleus of an atom. Nuclear fission provides the energy in nuclear power plants and fusion is the source of the sun's energy. The use of fission in power plants can help conserve fossil fuels. Without the energy produced by the fusion of hydrogen in the sun, the Earth would quickly change into a cold planet that could not support life as we know it.

Learning Objectives

- Compare and contrast nuclear fission and nuclear fusion.
- Understand the conservation laws that apply to nuclear reactions.

Success Criteria

- Find missing reactants/products in nuclear equations.
- Identify nuclear reactions as examples of fission or fusion.

Vocabulary

- Fission
- Fusion
- Transmutation
- Nuclide

Prerequisites

- Structure of atoms and isotopes.
- Knowledge of mass number and atomic number.
- Atomic symbols including mass number, atomic number, and charge.

Information

Transmutation is the transformation of the nucleus of an atom so that the atom is changed from one element into a different element. This can be accomplished through many types of reactions, including fission and fusion.

Nuclear transformations always obey two fundamental conservation laws:

(1) mass number is conserved and (2) electrical charge is conserved.

Energy and mass are not conserved, but can be inter-converted according to Einstein's equation, $E = mc^2$.

Nuclear Chemistry Pogil

Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami

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