Nuclear Chemistry Webquest

Name:

In this webauest, you will explore nuclear chemistry in real-world situations. You will learn about fusion and fission, types of radiation, its effects on humans, and how nuclear power is produced as well as its repercussions and disasters. Follow the steps below.

- Enter the following web address: www.resdk12.org/Page/37614
- 2. Under web, click on Nuclear Chemistry to open a copy of this document
- 3. Click on the links provided to answer the questions in each section

		Physics

- is the smallest particle of matter that maintains the properties of that element.
- 2. Sir Ernst Rutherford concluded what two things about atomic structure:
- Niels Bohr contributed what to atomic structure theory:
- 4. What is the unit of measure for the diameter of atoms?
- 5. What is the unit of measure from above equal to?
- 6. An atom consists of three basic subatomic particles. List each below. Include its charge and who discovered it.
- 7. The number of which subatomic particles determines the chemical element that it
- 8. The number and arrangement of which subatomic particle determines most of the atom's properties?
- 9. Protons and neutrons are composed of even smaller subatomic particles called
- 10. Define isotopes:

 11. Isotopes have the same _______ properties but very different properties.
- 12. Most isotopes are stable but some are
- 13. What is binding energy?
- 14. What does binding energy determine?
- 15. Einstein's famous equation of relativity, $E = mc^2$, explains the relationship between the binding energy and mass defect. It explains that a small amount of can produce a large amount of ______
- 16. Define radioactivity:
- 17. Name the three scientists that discovered and researched radioactivity:

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Patrick Vollmar

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Principles Of Nuclear Chemistry Peter A C Mcpherson, 2016-12-21 Principles of Nuclear Chemistry is an introductory text in nuclear chemistry and radiochemistry aimed at undergraduates with little or no knowledge of physics It covers the key aspects of modern nuclear chemistry and includes worked solutions to end of chapter questions. The text begins with basic theories in contemporary physics and uses these to introduce some fundamental mathematical techniques It relates nuclear phenomena to key divisions of chemistry such as atomic structure spectroscopy equilibria and kinetics It also gives an introduction to f block chemistry and the nuclear power industry. This book is essential reading for those taking a first course in nuclear chemistry and is a useful companion to other volumes in physical and analytical chemistry. It will also be of use to those new to working in nuclear chemistry or radiochemistry Radiochemistry and Nuclear Chemistry Gregory Choppin, Jan Rydberg, Jan-Olov Liljenzin, 2001-12-03 Radiochemistry or Nuclear Chemistry is the study of radiation from an atomic or molecular perspective including elemental transformation and reaction effects as well as physical health and medical properties This revised edition of one of the earliest and best known books on the subject has been updated to bring into teaching the latest developments in research and the current hot topics in the field In order to further enhance the functionality of this text the authors have added numerous teaching aids that include an interactive website that features testing examples in MathCAD with variable quantities and options hotlinks to relevant text sections from the book and online self grading texts As in the previous edition readers can closely follow the structure of the chapters from the broad introduction through the more in depth descriptions of radiochemistry then nuclear radiation chemistry and finally the guide to nuclear energy including energy production fuel cycle and waste management New edition of a well known respected text in the specialized field of nuclear radiochemistry Includes an interactive website with testing and evaluation modules based on exercises in the book Suitable for both radiochemistry and nuclear chemistry courses **Nuclear Chemistry** Maheshwar Sharon, Madhuri Sharon, 2021-02-01 This book is designed to serve as a textbook for core courses offered to postgraduate students enrolled in chemistry This book can also be used as a core or supplementary text for nuclear chemistry courses offered to students of chemical engineering The book covers various topics of nuclear chemistry like Shell model fission fusion reaction natural radioactive equilibrium series nuclear reactions carried by various types of accelerators In addition it describes the law of decay of radioactivity type of decay and interaction of radiation with matter It explains the difference between ionization counter scintillation counter and solid state detector This book also consists of end of book problems to help readers aid self learning The detailed coverage and pedagogical tools make this an ideal textbook for postgraduate students and researchers enrolled in various chemistry and engineering courses This book will also be beneficial for industry professionals in the allied fields Nuclear Chemistry ,2010 **Radiochemistry and Nuclear** Chemistry Gregory Choppin, Jan-Olov Liljenzin, Jan Rydberg, Christian Ekberg, 2013-09-05 Radiochemistry or nuclear

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Nuclear Chemistry California. University. Lawrence Radiation Laboratory. Nuclear Chemistry Division, 1968

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