

Name: _____

Nuclear Chemistry Worksheet #1

Complete the following nuclear equations:

1.
$$\begin{array}{c} 187 \\ 75 \end{array} \text{Re} + \underline{\hspace{2cm}} \rightarrow \begin{array}{c} 188 \\ 75 \end{array} \text{Re} + \begin{array}{c} 1 \\ 1 \end{array} \text{H}$$
2.
$$\begin{array}{c} 9 \\ 4 \end{array} \text{Be} + \begin{array}{c} 4 \\ 2 \end{array} \text{He} \rightarrow \begin{array}{c} 1 \\ 0 \end{array} \text{n} + \underline{\hspace{2cm}}$$
3.
$$\begin{array}{c} 22 \\ 11 \end{array} \text{Na} + \underline{\hspace{2cm}} \rightarrow \begin{array}{c} 22 \\ 10 \end{array} \text{Ne}$$
4.
$$\begin{array}{c} 238 \\ 93 \end{array} \text{Np} \rightarrow \begin{array}{c} 4 \\ 2 \end{array} \text{He} + \underline{\hspace{2cm}}$$

Complete the following nuclear reactions and state the type of radioactive decay:

5.
$$\begin{array}{c} 202 \\ 80 \end{array} \text{Hg} \rightarrow \begin{array}{c} 198 \\ 78 \end{array} \text{Pt} + \underline{\hspace{2cm}}$$
6.
$$\begin{array}{c} 232 \\ 91 \end{array} \text{Pa} \rightarrow \underline{\hspace{2cm}} + \begin{array}{c} 0 \\ -1 \end{array} \beta$$
7.
$$\begin{array}{c} 46 \\ 22 \end{array} \text{Ti} \rightarrow \begin{array}{c} 46 \\ 21 \end{array} \text{Sc} + \underline{\hspace{2cm}}$$

8. Write an equation to represent the decay of curium-247 by alpha emission.

Answer the following questions.

9. The splitting of the nucleus into lighter nuclei is called _____.
10. When light mass nuclei combine to form a heavier, more stable nucleus this is called _____.
11. Do current nuclear power plants "harvest" energy from fusion or fission reactions? Why?
12. List two examples of fission reactions:
13. List two examples of fusion reactions:
14. List four types of nuclear waste:

Nuclear Chemistry Worksheet Section 2 Answers

Michael Brown



Nuclear Chemistry Worksheet Section 2 Answers:

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