

Balancing Nuclear Equations

Name: _____ Date: _____ Period: _____

Balancing Nuclear Equations

Directions: For each of the following nuclear equations insert the correct number of electrons (e^-), neutrons (n), or other isotopes. Remember you can add coefficients to your answer.

1. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C}$	16. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
2. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C}$	17. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
3. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	18. $\text{_____} + ^{12}_6\text{C} \rightarrow ^{12}_6\text{C} + ^{12}_6\text{C}$
4. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C}$	19. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C}$
5. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	20. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
6. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	21. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
7. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	22. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
8. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	23. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
9. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C}$	24. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
10. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	25. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
11. $^{12}_6\text{C} + \text{_____} + ^{12}_6\text{C} \rightarrow \text{_____}$	26. $^{12}_6\text{C} + \text{_____} \rightarrow ^{12}_6\text{C} + ^{12}_6\text{C}$
12. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	27. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
13. $^{12}_6\text{C} + ^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$	28. $^{12}_6\text{C} + ^{12}_6\text{C} \rightarrow \text{_____}$
14. $^{12}_6\text{C} + \text{_____} + ^{12}_6\text{C} \rightarrow \text{_____}$	29. $^{12}_6\text{C} + \text{_____} + ^{12}_6\text{C} \rightarrow \text{_____}$
15. $^{12}_6\text{C} + \text{_____} + ^{12}_6\text{C} \rightarrow \text{_____}$	30. $^{12}_6\text{C} + \text{_____} + ^{12}_6\text{C} \rightarrow \text{_____}$

- 1-Page worksheet with Answer Key
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