

Challenge Nuclear Decay Problem Set

Directions: Using the periodic table and your knowledge of nuclear decay, identify the missing atomic nuclei or radiation particles in the following nuclear equations....

1. Alpha decay of radium-226, the most abundant isotope of radium



2. Radioactive decay of carbon-14, which is used in radiocarbon dating



3. "Electron capture" by potassium-40, a natural source of radiation in the human body



4. Alpha decay of the artificially produced radioisotope that is used in smoke detectors



5. Formation of radioactive carbon-14 in the upper atmosphere by reaction with cosmic rays



6. Beta decay of iodine-131, which is used to treat thyroid cancer



Nuclear Equations Radioactive Decay Answer Key

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