



Nuclear Radiation Worksheet

a. What determines whether an atom is stable or unstable?

It's overall size and a difference between proper it and prentom it

b. How does an unstable atom gain stability?

It undergoes redisastive decay

c. What part of an atom undergoes change during radioactive decay?

The Mucheus !

Which of the human senses can detect radiation?

ich of the human senses can detect radiation?
Our eyes can detect colors, but we have no senses to detect alpha, between patriotion

Explain the difference between an isotope and a radioactive isotope.
 Isotope has a defend that payou feeders. Radiodox 15 league modelle and has a defend at 4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	The state of	Positive C+1	Storped by files in proper
Beta	O No Ress (Garren L.)	Directive (+)	compathy duch of his full or this impact
Gamma	B. No mark	Nambral	Should be sever when it had or fact

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No chestre	41
Gamma	Un akenny	No change

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.

Example: ${}^{218}_{84}Po \rightarrow {}^{4}_{2}He + {}^{214}_{82}Pb$

a.
$$^{210}Pb \rightarrow ^{610}_{83}Bi + ^{6}_{-1}e^{-\beta}$$

b.
$$^{230}Th \rightarrow ^{17}Ra + ^{4}He$$

- 7. Complete the following nuclear equations:
- $^{239}_{83}Bi \rightarrow ^{4}_{2}He + ^{235}_{21}\mathcal{T}$ b. 239 Np → 239 Pu + -1

Nuclear Radiation Answer Key Chemistry

L Cohen

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Chemical Principles Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding Hearings United States. Congress. House. Committee on Appropriations, 1956 States. Congress. House, 1969 Second Supplemental Appropriation Bill, 1957 United States. Congress. House. Committee on Appropriations, 1956 Report on Foreign Policy and Mutual Security United States. Congress. House. Committee on Foreign Affairs, 1957 Foreign Policy and Mutual Security United States. Congress. House. Committee on Foreign *Public Works Appropriations for 1970 for Water and Power Resources* Affairs, 1956 Energy Research Abstracts ,1993 Development and the Atomic Energy Commission United States, Congress, House, Committee on Appropriations, 1969

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