

MULTIPLE CHOICE QUESTIONS

Oxidation Number

- Oxidation is a process of
- loss of electrons
 - gain of electrons
 - increase in the negative valency
 - decrease in positive valency.
- When zinc is added to CuSO_4 solution, copper is precipitated, it is so because of
- Reduction of Zn
 - Reduction of Cu^{2+}
 - Hydrolysis of CuSO_4
 - Reduction of SO_4^{2-} .
- The process in which oxidation number increases is
- Reduction
 - Hydrolysis
 - Oxidation
 - Decomposition.
- When zinc metal is added to dilute H_2SO_4 solution, hydrogen is evolved. In this zinc undergoes
- Reduction
 - Oxidation
 - Dissolution
 - None of the above.
- In which of the following, oxidation number of chlorine is +5 ?
- Cl_2O_7
 - ClO_3^-
 - ClO^-
 - ClO_4^- .
- Carbon has zero oxidation number in
- CH_4
 - CH_3Cl
 - CCl_4
 - CH_2Cl_2 .
- The oxidation number of carbon in $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ is
- 0
 - +22
 - +6
 - 6.
- In the following
- $$3\text{Br}_2 + 6\text{CO}_3^{2-} + 3\text{H}_2\text{O} \longrightarrow 5\text{Br}^- + \text{BrO}_3^- + 6\text{HCO}_3^-$$
- Bromine is oxidised and carbonate is reduced
 - Bromine is reduced and carbonate is oxidised

- Bromine is neither reduced nor oxidised
 - Bromine is reduced and oxidised.
9. The oxidation number of C in CH_4 , CH_3Cl , CH_2Cl_2 , CHCl_3 and CCl_4 is respectively
- 4, -2, 0, +2, +4
 - +2, 4, 0, -2, -4
 - 4, 2, 0, -2, -4
 - 0, 2, -2, 4, 4.
10. Chlorine is in +3 oxidation state in
- HCl
 - HClO_4
 - ICl
 - ClF_3 .
11. Oxidation number of Mn in K_2MnO_4 is
- 2
 - 4
 - 6
 - 7.
12. A student states that burning of lime in air is an oxidation process. The reason he gives is that an oxide of the metal is produced on burning. Which one is correct ?
- The statement and reason are true, the reason is correct explanation
 - The statement and reason are true, the explanation is not correct
 - The statement is true but the reason is false
 - The statement is false but the reason is true.
13. For the following reaction in the acidic solution
- $$\text{MnO}_4^- + 8\text{H}^+ + 5e^- \longrightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O},$$
- which of the following gives the true oxidation numbers of the manganese on each side of the equation ?
- +7 to +6
 - +7 to +2
 - +4 to +2
 - 1 to +2.
14. In which of the following compounds, iron has the lowest oxidation number ?
- $\text{Fe}(\text{CO})_5$
 - Fe_2O_3
 - $\text{K}_4[\text{Fe}(\text{CN})_6]$
 - $\text{FeSO}_4(\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$.
15. The oxidation number and covalency of sulphur in sulphur molecule (S_8) are

Answers

- | | | | | | | | | | |
|---------|---------|---------|---------|--------|--------|--------|--------|--------|---------|
| 1. (A) | 2. (B) | 3. (C) | 4. (B) | 5. (B) | 6. (D) | 7. (A) | 8. (D) | 9. (A) | 10. (D) |
| 11. (C) | 12. (D) | 13. (B) | 14. (A) | | | | | | |

Redox Multiple Choice Questions And Answers

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Table of Contents Redox Multiple Choice Questions And Answers

1. Understanding the eBook Redox Multiple Choice Questions And Answers
 - The Rise of Digital Reading Redox Multiple Choice Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Redox Multiple Choice Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Redox Multiple Choice Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Redox Multiple Choice Questions And Answers
 - Personalized Recommendations
 - Redox Multiple Choice Questions And Answers User Reviews and Ratings
 - Redox Multiple Choice Questions And Answers and Bestseller Lists

5. Accessing Redox Multiple Choice Questions And Answers Free and Paid eBooks
 - Redox Multiple Choice Questions And Answers Public Domain eBooks
 - Redox Multiple Choice Questions And Answers eBook Subscription Services
 - Redox Multiple Choice Questions And Answers Budget-Friendly Options
6. Navigating Redox Multiple Choice Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Redox Multiple Choice Questions And Answers Compatibility with Devices
 - Redox Multiple Choice Questions And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Redox Multiple Choice Questions And Answers
 - Highlighting and Note-Taking Redox Multiple Choice Questions And Answers
 - Interactive Elements Redox Multiple Choice Questions And Answers
8. Staying Engaged with Redox Multiple Choice Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Redox Multiple Choice Questions And Answers
9. Balancing eBooks and Physical Books Redox Multiple Choice Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Redox Multiple Choice Questions And Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Redox Multiple Choice Questions And Answers
 - Setting Reading Goals Redox Multiple Choice Questions And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Redox Multiple Choice Questions And Answers
 - Fact-Checking eBook Content of Redox Multiple Choice Questions And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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