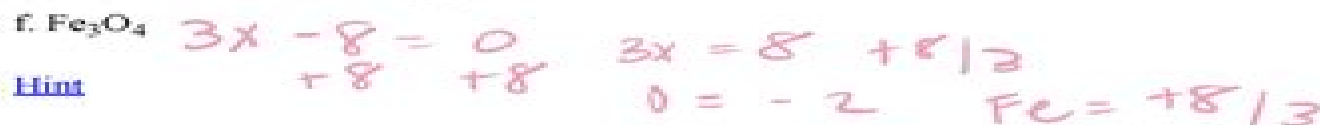


Practice Problems: Redox Reactions

1. Determine the oxidation number of the elements in each of the following compounds:



Hint

2. Identify the species being oxidized and reduced in each of the following reactions:



3. Would you use an oxidizing agent or reducing agent in order for the following reactions to occur?



Hint

Redox Practice Problems With Answers

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Redox Practice Problems With Answers:

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

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Chemical Literacy and Writing Chemical Reactions Nikolay Gerasimchuk, 2022-09-12

Writing chemical reactions in general and inorganic chemistry is not a trivial task. However, writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject. Unfortunately, very few students grasp the concept of the correct writing of chemical reactions quickly and so are unable to move through topics of general analytical and inorganic chemistry freely. Because the ability to write and balance different types of chemical reactions is a fundamental issue, this becomes a key question of chemical literacy. The successful writing of chemical reactions includes two components: the prediction of products of these reactions and their possible variations and balancing these reactions, providing a material balance between starting compounds and reactions products. This book explores that element of the teaching of the fundamentals of chemical literacy: writing complete equations of chemical reactions and balancing them. It contains 49 figures, 22 schemes and 12 tables and 93 problems with answers. This book will be very useful for high school students interested in chemical sciences, higher education teachers, students in colleges and universities majoring in chemistry and biochemistry, and chemistry professionals working in industry. It also contains information about properties of the most

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