



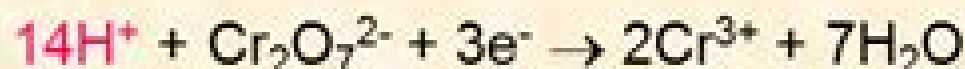
Step 2: Balance each kind of atom other than H and O



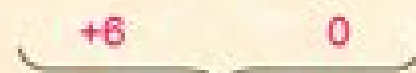
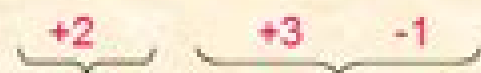
Step 3: Balance oxygen atoms by using  $\text{H}_2\text{O}$



Step 4: Balance H atoms by using  $\text{H}^+$  ions



Step 5: Use electrons as needed to obtain a charge that is balanced



+2

+2

+9

+6

Add three electrons to the reactant side to balance the charges



# Redox Questions And Answers

**S. Chand Experts**



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Biochemistry Theory and Practicals Questions and Answers Mr. Rohit Manglik, 2024-07-24 A comprehensive Q A resource that prepares students for exams and lab work in biochemistry through concise theoretical explanations and practical experiment guidance

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**Chemistry-vol-II** Dr Vijay Sarda, Dr A C Handa, Dr K K Arora, A text book on Chemistry

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