

1a. i. Magnesium + Oxygen $\longrightarrow$ Magnesium Oxide

ii. Exothermic reaction.

b. i. "C" Oxygen.

ii. As Oxygen supports combustion and will burn the magnesium brightly.

iii. "A".

iv. as it contains Exhaled air that consists of carbon dioxide which does not support combustion.

c. i. Gold

ii. Used in jewelry and ornaments.

iii. As gold is inert that means it does not react with oxygen or water. It keeps its lustre.

2a. Acids react with limestone. Acids turn the universal indicator red. Acids damage teeth. Acids may harm your skin. Acids react with metals such as magnesium.

b. When metals react with acids the gas produced is hydrogen.

When this gas is burned it is oxidised.

The word equation for the reaction when this gas burns is Hydrogen + Oxygen $\longrightarrow$ Water.

Write word equations between the following with hydrochloric acid and then water:

i. Calcium + hydrochloric acid $\longrightarrow$ Calcium chloride + hydrogen

Calcium + water $\longrightarrow$ Calcium hydroxide + hydrogen

ii. Magnesium + hydrochloric acid $\longrightarrow$ Magnesium chloride + hydrogen

Magnesium + water $\longrightarrow$ Magnesium hydroxide + hydrogen.

ii. Iron + hydrochloric acid $\longrightarrow$ Iron chloride + hydrogen.

Iron + water $\longrightarrow$ Iron hydroxide + hydrogen.

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Xiang Xie



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