# 5.3 - Periodic Trends

Name	
I. Short answer	
<ol> <li>Label each group on the periodic table.</li> </ol>	
<ol><li>Explain what valence electrons are.</li></ol>	
3. Explain what makes an element stable?	
II. Answer the questions using the symbols on the	periodic table
1. What elements are in period 5?	•
What element(s) is a	
2.halogens	6transition metals
3. lanthanoids	7alkali metals
4alkaline earth metals	8.actinoids
5noble gases	
What elements are in the sublevel	
9 s	11. d
10 p	12. f
What elements are	
13. metals,	
14, nonmetals	
15.metalloids,	
16. Which element is the most reactive no	nmetal?
17. Which element is the most reactive m	etal?
18.Which element below is the least react	ive?
19. Which element has the largest radius	in period 4?
20. Which element has the largest radius in	n the Alkali metals family?
21. Which element has the highest ionizate	ion energy in period four?
22. Which elements has the second largest	atomic number?
23. Which element has the largest atomic	mass?
24. Which elements outer electron config	uration is 5s1?
25. Which elements outer electron config	uration is 7s <sup>2</sup> 5t <sup>6</sup> ?
27 Which elements will form +1 ions?	
28. Which group of metals will have the sa	me properties
29 Which group of nonmetals will have th	e same properties
30. On the back, define all chapter 5, section	on 3 vocabulary terms.

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i			t.					
		c				z		

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**Puja Mehta** 

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#### Table of Contents Periodic Puzzle Worksheet Answers 2014

- 1. Understanding the eBook Periodic Puzzle Worksheet Answers 2014
  - The Rise of Digital Reading Periodic Puzzle Worksheet Answers 2014
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Puzzle Worksheet Answers 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Puzzle Worksheet Answers 2014
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Puzzle Worksheet Answers 2014
  - Personalized Recommendations
  - Periodic Puzzle Worksheet Answers 2014 User Reviews and Ratings
  - Periodic Puzzle Worksheet Answers 2014 and Bestseller Lists
- 5. Accessing Periodic Puzzle Worksheet Answers 2014 Free and Paid eBooks

- Periodic Puzzle Worksheet Answers 2014 Public Domain eBooks
- Periodic Puzzle Worksheet Answers 2014 eBook Subscription Services
- Periodic Puzzle Worksheet Answers 2014 Budget-Friendly Options
- 6. Navigating Periodic Puzzle Worksheet Answers 2014 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Periodic Puzzle Worksheet Answers 2014 Compatibility with Devices
  - Periodic Puzzle Worksheet Answers 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Puzzle Worksheet Answers 2014
  - Highlighting and Note-Taking Periodic Puzzle Worksheet Answers 2014
  - Interactive Elements Periodic Puzzle Worksheet Answers 2014
- 8. Staying Engaged with Periodic Puzzle Worksheet Answers 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Puzzle Worksheet Answers 2014
- 9. Balancing eBooks and Physical Books Periodic Puzzle Worksheet Answers 2014
  - Benefits of a Digital Library
  - $\circ$  Creating a Diverse Reading Collection Periodic Puzzle Worksheet Answers 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Puzzle Worksheet Answers 2014
  - Setting Reading Goals Periodic Puzzle Worksheet Answers 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Puzzle Worksheet Answers 2014
  - Fact-Checking eBook Content of Periodic Puzzle Worksheet Answers 2014
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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