

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The image shows a crossword puzzle grid with letters placed in specific cells. The letters are as follows:

- Row 1:** A (1,1), B (1,14)
- Row 2:** C (2,1), D (2,11), E (2,13)
- Row 3:** F (3,2), G (3,7), H (3,11), K (3,14)
- Row 4:** L (4,2), I (4,13)
- Row 5:** J (5,1), Z (5,3), M (5,7), N (5,11), O (5,13), P (5,14)
- Row 6:** Q (6,1), R (6,3), S (6,7), T (6,8), U (6,9)

Below the main grid, there is a separate row of cells. Arrows point from the letter 'R' in the main grid to the starting cells of two words in this row:

- Word 1: V (7,1), W (7,10)
- Word 2: X (7,1), Y (7,10)

List elements in *alphabetical* order when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas    **B)** it is a gas    **C)** it is NOT a gas    **D)** it does NOT share the properties of the other elements in its group    **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C"    **B)** "E"    **C)** "K"    **D)** A, B, and C    **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "E"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "D"    **B)** "Y"    **C)** "A"    **D)** "O"    **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C"    **B)** "N"    **C)** "H"    **D)** "K"    **E)** "Y"

continued on other side

# Periodic Table Packet Graphs Worksheets Answers

**Desmond W. Evans**



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