

ATOMIC MASS INCREASES

The Periodic Table of the Elements is displayed in a grid format. The horizontal rows are labeled with element numbers 1 through 7. The vertical columns are labeled with group numbers 1A through 18VIA. The table includes the following features:

- Electronegativity Increase:** Indicated by a green downward-pointing arrow on the left side of the table.
- Atomic Mass Increases:** Indicated by a blue upward-pointing arrow at the top of the table.
- Atomic Mass Increases:** Indicated by a blue downward-pointing arrow on the right side of the table.
- Electronegativity Increase:** Indicated by a green upward-pointing arrow at the bottom of the table.
- Color Coding:** Elements are color-coded based on their properties:
 - Green:** Hydrogen (H), Helium (He), Fluorine (F), Oxygen (O), Nitrogen (N), Carbon (C), Boron (B), and Hydrogen-like elements (Li, Be, Na, Mg, K, Ca, Rb, Sr, Cs, Ba, Fr, Ra).
 - Pink:** Lithium (Li), Potassium (K), Rubidium (Rb), Cesium (Cs), and Francium (Fr).
 - Blue:** Boron (B), Nitrogen (N), Oxygen (O), Fluorine (F), Neon (Ne), Argon (Ar), Krypton (Kr), Xenon (Xe), and Radon (Rn).
 - Yellow:** Helium (He), Hydrogen (H), Boron (B), Carbon (C), Nitrogen (N), Oxygen (O), Fluorine (F), Chlorine (Cl), Sulfur (S), Phosphorus (P), Silicon (Si), Germanium (Ge), Arsenic (As), Antimony (Sb), Tellurium (Te), Iodine (I), and the lanthanides (La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu) and actinides (Ac, Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, No, Lr).
 - Orange:** Hydrogen (H), Helium (He), Boron (B), Carbon (C), Nitrogen (N), Oxygen (O), Fluorine (F), Chlorine (Cl), Sulfur (S), Phosphorus (P), Silicon (Si), Germanium (Ge), Arsenic (As), Antimony (Sb), Tellurium (Te), Iodine (I), and the transition metals (Cr, Mn, Fe, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Br, Kr, In, Sn, Sb, Te, I, Po, At, Rn).

ELECTRONEGATIVITY INCREASE

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Understanding Advanced Physical Inorganic Chemistry: The Learner's Approach (Revised Edition) Kim Seng Chan,Jeanne Tan,2016-09-26 This revised edition has been updated to meet the minimum requirements of the new Singapore GCE A level syllabus that would be implemented in the year 2016 Nevertheless this book is also highly relevant to students who are studying chemistry for other examination boards In addition the authors have also included more Q A to help students better understand and appreciate the chemical concepts that they are mastering

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