

Periodic Table Quiz

hydrogen 1 H 1.00794																		helium 2 He 4.002602					
lithium 3 Li 6.941	beryllium 4 Be 9.012182																	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	aluminum 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.06	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078																	gallium 31 Ga 69.723	germanium 32 Ge 72.64	arsenic 33 As 74.922	selecnium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80
rubidium 37 Rb 85.468	strontium 38 Sr 87.62																	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29
cesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lanthanum 71 Lu 138.91	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po 209	astatine 85 At 210	radon 86 Rn 222					
francium 87 Fr 223	radium 88 Ra 226	89-102 **	actinium 89 Lr 227	thorium 90 Rf 232	protactinium 91 Db 231	uranium 92 Sg 238	neptunium 93 Bh 237	plutonium 94 Hs 244	americium 95 Mt 243	curium 96 Uun 247	berkelium 97 Uuu 247	californium 98 Uub 251					einsteinium 99 Uuq 252						

*lanthanoids

**actinoids

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm 145	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.05
actinium 89 Ac 227	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np 237.05	plutonium 94 Pu 244	americium 95 Am 243	curium 96 Cm 247	berkelium 97 Bk 247	californium 98 Cf 251	einsteinium 99 Es 252	fermium 100 Fm 257	mendelevium 101 Md 258	nobelium 102 No 259

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Theodore S. Abbgy



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thermal decomposition of group II carbonates thermal decomposition of group II nitrates uses of group II elements uses of group II metals uses of halogens and their compounds The Halogenoalkanes MCQ PDF e Book Chapter 14 practice test to solve MCQ questions on Halogenoalkanes uses of halogenoalkanes elimination reactions nucleophilic substitution in halogenoalkanes and nucleophilic substitution reactions The Hydrocarbons MCQ PDF e Book Chapter 15 practice test to solve MCQ questions on Introduction to alkanes sources of alkanes addition reactions of alkenes alkane reaction alkenes and formulas The Introduction to Organic Chemistry MCQ PDF e Book Chapter 16 practice test to solve MCQ questions on Organic chemistry functional groups organic reactions naming organic compounds stereoisomerism structural isomerism and types of organic reactions The Ionic Equilibria MCQ PDF e Book Chapter 17 practice test to solve MCQ questions on Introduction to ionic equilibria buffer solutions equilibrium and solubility indicators and acid base titrations pH calculations and weak acids The Lattice Energy MCQ PDF e Book Chapter 18 practice test to solve MCQ questions on Introduction to lattice energy ion polarization lattice energy value atomization and electron affinity Born Haber cycle and enthalpy changes in solution The Moles and Equations MCQ PDF e Book Chapter 19 practice test to solve MCQ questions on Amount of substance atoms molecules mass chemical formula and equations gas volumes mole calculations relative atomic mass solutions and concentrations The Nitrogen and Sulfur MCQ PDF e Book Chapter 20 practice test to solve MCQ questions on Nitrogen gas nitrogen and its compounds nitrogen and gas properties ammonia ammonium compounds environmental problems caused by nitrogen compounds and nitrate fertilizers sulfur and oxides sulfuric acid and properties and uses of sulfuric acid The Organic and Nitrogen Compounds MCQ PDF e Book Chapter 21 practice test to solve MCQ questions on Amides in chemistry amines amino acids peptides and proteins The Periodicity MCQ PDF e Book Chapter 22 practice test to solve MCQ questions on Acidic oxides basic oxides aluminum oxide balancing equation period 3 chlorides balancing equations reactions with chlorine balancing equations reactions with oxygen bonding nature of period 3 oxides chemical properties of chlorine chemical properties of oxygen chemical properties periodicity chemistry periodic table chemistry oxides chlorides of period 3 elements electrical conductivity in period 3 oxides electronegativity of period 3 oxides ionic bonds molecular structures of period 3 oxides oxidation number of oxides oxidation numbers oxides and hydroxides of period 3 elements oxides of period 3 elements period III chlorides periodic table electronegativity physical properties periodicity reaction of sodium and magnesium with water and relative melting point of period 3 oxides The Polymerization MCQ PDF e Book Chapter 23 practice test to solve MCQ questions on Types of polymerization polyamides polyesters and polymer deductions The Rates of Reaction MCQ PDF e Book Chapter 24 practice test to solve MCQ questions on Catalysis collision theory effect of concentration reaction kinetics and temperature effect on reaction rate The Reaction Kinetics MCQ PDF e Book Chapter 25 practice test to solve MCQ questions on Reaction kinetics catalysts kinetics and reaction mechanism order of reaction rate constant k and rate of reaction The Redox Reactions and Electrolysis MCQ PDF e Book Chapter 26 practice

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