

## Periodic Table WebQuest

**Periodic Table of the Elements**

**Lanthanide Series**

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
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**Actinide Series**

Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
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**Directions: In each of the websites listed, find the information asked for.**

Use <http://atomictimeline.net/index.php> to write the contributions of each person toward the development of their theory of the atom (atomic model) and the Periodic Table of Elements.

1. Democritus
2. John Dalton
3. J. J. Thompson
4. Niels Bohr
5. Ernest Rutherford

Use [http://www.kyrene.org/Staff/mijohns/atomic\\_structure\\_website/electroncloud.htm](http://www.kyrene.org/Staff/mijohns/atomic_structure_website/electroncloud.htm), [http://education.jlab.org/qa/particlemass\\_02.html](http://education.jlab.org/qa/particlemass_02.html), and <http://www.colorado.edu/physics/2000/isotopes/index.html>, to answer the following:

6. List 3 unique aspects of the electron cloud model in general comparison to previous models.
7. How does the mass of a proton compare to the mass of a neutron?
8. How does the mass of a proton compare to the mass of an electron?

# Periodicity Webquest Answer Key

**Katrin Zwirglmaier**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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