#### Lesson Quiz

# **Observing Chemical Change**

Write the letter of the correct answer on the line at the left.

- Which of the following is true about chemical reactions?
  - A They are accompanied by changes in energy.
  - B They form new substances with new properties.
  - C both A and B
  - D neither A nor B
- Which of the following is NOT a physical property?
  - A melting point
  - B state of matter
  - C density
  - D flammability

- 2. \_\_\_ In an endothermic reaction, energy is
  - A absorbed
  - B released
  - C converted to mass
  - D synthesized
- Substances formed as a result of a chemical reaction are called
  - A catalysts
  - B precipitates
  - C products
  - D reactants

If the statement is true, write true. If the statement is false, change the underlined word or words to make the statement true.

- In an exothermic reaction, products have more energy than reactants.
- Water boils at 100°C. This is an example of a chemical property.
- Substances that enter into a chemical reaction are called products.
- The ability to react with oxygen is an example of a <u>chemical</u> property.
- Another name for a chemical change is a chemical bond.
- 10. \_\_\_\_\_ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

**Victor M. Corman** 

Focus on Physical Science California Edition Michael J. Padilla, 2007 **Investigations in Natural Science: pt.1.** Chemistry. Teacher's guide ,1985 Science Educator's Guide to Laboratory Assessment Rodney L. Doran, 2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during Matter and Energy .2005 **Resources in Education** .1998 The Guide to Learning and Study Skills science lab Ms Rosie Bingham, Ms Sue Drew, 2012-04-01 This new guide builds on the hugely successful materials the authors have developed over the last 15 years Along with highly practical guidance on traditional learning skills The Guide to Learning and Study Skills provides guidance for students on learning in a blended environment the increased use of personal and professional development planning continuing professional development and work based learning Psychology of Reading Keith Rayner, Alexander Pollatsek, Jane Ashby, Charles Clifton Literary Chronicle ,1837 Jr., 2012-05-22 Reading is a highly complex skill that is prerequisite to success in many societies in which a great deal of information is communicated in written form Since the 1970s much has been learned about the reading process from research by cognitive psychologists This book summarizes that important work and puts it into a coherent framework The book s central theme is how readers go about extracting information from the printed page and comprehending the text Like its predecessor this thoroughly updated 2nd Edition encompasses all aspects of the psychology of reading with chapters on writing systems word recognition the work of the eyes during reading inner speech sentence processing discourse processing learning to read dyslexia individual differences and speed reading Psychology of Reading 2nd Edition is essential reading for undergraduates graduates and researchers in cognitive psychology and could be used as a core textbook on courses on the psychology of reading and related topics In addition the clear writing style makes the book accessible to people without a background in psychology but who have a personal or professional interest in the process of reading Scientific American ,1883 Biography-Driven Culturally Responsive Teaching, Second Edition Socorro G. Herrera, 2016-01-27 Culturally responsive pedagogy literacy and English learner education expert Socorro Herrera has updated this bestseller to clarify focus and redefine concepts for the continued professional development of educators serving culturally and linguistically diverse CLD populations Teaching strategies and tools have been updated to reflect important new brain research and to keep pace with our nation's ever changing demographics and constant shift in expectations for K 12 students Herrera has also revised the structure and format of the book to help educators find information quickly while working in highly complex and demanding environments New for the Second Edition Teaching strategies and tools based on the most current knowledge in the field Authentic classroom artifacts that have been collected from teachers across the country Glossary of

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Surgical Journal ,1893 Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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