

# PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



## Basic Information

Name: Hydrogen  
Symbol: H  
Atomic Number: 1  
Atomic Mass: 1.00794 amu  
Melting Point:  $-259.14^{\circ}\text{C}$  ( $-314.59995^{\circ}\text{K}$ ,  $-434.41203^{\circ}\text{F}$ )  
Boiling Point:  $-252.87^{\circ}\text{C}$  ( $-335.35595^{\circ}\text{K}$ ,  $-425.164^{\circ}\text{F}$ )  
Number of Protons/Electrons: 1  
Number of Neutrons: 0  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @  $293^{\circ}\text{K}$ :  $0.08989\text{ g/cm}^3$   
Color: colorless

## Atomic Structure

Number of Energy Levels: 1  
First Energy Level: 1

## Isotopes

Isotope	Half Life
H-1	Stable
H-2	Stable
H-3	12.5 years

## Facts

Name Origin: Greek words *hydro* (water) and *genes* (generated)  
Uses: Balloons, metal refining  
Obtained From: natural, all gas water

Molecular Structure of Carbon



## Name: Carbon

Symbol: C  
Atomic Number: 6  
Atomic Mass: 12.0097 amu  
Melting Point:  $3500^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6312.8^{\circ}\text{F}$ )  
Boiling Point:  $4602.18^{\circ}\text{C}$  ( $5100.13^{\circ}\text{K}$ ,  $8320.9^{\circ}\text{F}$ )  
Number of Protons/Electrons: 6  
Number of Neutrons: 6  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @  $293^{\circ}\text{K}$ :  $2.62\text{ g/cm}^3$   
Color: May be black

## Atomic Structure

Number of Energy Levels: 2  
First Energy Level: 2      Second Energy Level: 4

## Isotopes

Isotope	Half Life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730 years
C-15	2.3 seconds

## Facts

Uses: steel, flame  
Obtained From: burning with insufficient oxygen

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