

# **STRUCTURE OF A REFLECTIVE ESSAY**

## **INTRODUCTION**

**Appealing and effective  
Must have argument, problem or debate**

## **BODY SECTION**

**Present evidence supporting your story  
Must have complete and linking sentences  
Share a proper analysis**

## **CONCLUSION**

**Brief of what you shared  
No new information  
Include a reflective question**

# Personal Reflective Essay Structure

**Barbara K. Hofer, Paul R. Pintrich**



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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation  $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$  now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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