

In the extension section of experiment #9b, a student found that absolute zero was equal to -273 degrees celsius. For that graph to be true, we need to make one assumption about gases (which is also one of the assumptions in the Kinetic Molecular Theory of gases) which is...

- ☐ a) The volume of a gas is negligible (or zero) at absolute zero.
- ☐ b) Gases move really fast at absolute zero.
- ☐ c) The identity of a gas must be known to calculate its pressure
- ☐ d) All gases have the same mass.

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Thomas Griffiths



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