

List of Common Polyatomic Ions

Cations:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
NH_4^+	Ammonium	+1	Hg_2^{+2}	Dimercury (I)	+2

Anions:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{C}_2\text{H}_3\text{O}_2^-$	Acetate	-1	CO_3^{-2}	Carbonate	-2
HCO_3^-	Bicarbonate	-1	CrO_4^{-2}	Chromate	-2
BrO_3^-	Bromate	-1	$\text{Cr}_2\text{O}_7^{-2}$	Dichromate	-2
BrO_2^-	Bromite	-1	$\text{C}_2\text{O}_4^{-2}$	Oxalate	-2
ClO_4^-	Perchlorate	-1	O_2^{-2}	Peroxide	-2
ClO_3^-	Chlorate	-1	SO_4^{-2}	Sulfate	-2
ClO_2^-	Chlorite	-1	SO_3^{-2}	Sulfite	-2
ClO^-	Hypochlorite	-1	$\text{S}_2\text{O}_3^{-2}$	Thiosulfate	-2
CN^-	Cyanide	-1	SiO_3^{-2}	Silicate	-2
SCN^-	Thiocyanate	-1	MoO_4^{-2}	Molybdate	-2
OH^-	Hydroxide	-1	WO_4^{-2}	Tungstate	-2
NO_3^-	Nitrate	-1	BO_3^{-3}	Borate	-3
NO_2^-	Nitrite	-1	PO_4^{-3}	Phosphate	-3
MnO_4^-	Permanganate	-1	PO_3^{-3}	Phosphite	-3
HSO_4^-	Bisulfate	-1	AsO_4^{-3}	Arsenate	-3
HSO_3^-	Bisulfite	-1	AsO_3^{-3}	Arsenite	-3
IO_3^-	Iodate	-1			
IO_4^-	Periodate	-1			
$\text{C}_6\text{H}_5\text{COO}^-$	Benzoate	-1			

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{H}_2\text{PO}_4^{-1}$	Dihydrogen phosphate	-1
HPO_4^{-2}	Hydrogen phosphate	-2
$\text{Fe}(\text{CN})_6^{-3}$	Hexacyanoferrate (III)	-3
$\text{Fe}(\text{CN})_6^{-4}$	Hexacyanoferrate (II)	-4

Polyatomic Ions Extension Questions Answers

M Carnoy



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Table of Contents Polyatomic Ions Extension Questions Answers

1. Understanding the eBook Polyatomic Ions Extension Questions Answers
 - The Rise of Digital Reading Polyatomic Ions Extension Questions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Extension Questions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ions Extension Questions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Extension Questions Answers
 - Personalized Recommendations
 - Polyatomic Ions Extension Questions Answers User Reviews and Ratings

- Polyatomic Ions Extension Questions Answers and Bestseller Lists
- 5. Accessing Polyatomic Ions Extension Questions Answers Free and Paid eBooks
 - Polyatomic Ions Extension Questions Answers Public Domain eBooks
 - Polyatomic Ions Extension Questions Answers eBook Subscription Services
 - Polyatomic Ions Extension Questions Answers Budget-Friendly Options
- 6. Navigating Polyatomic Ions Extension Questions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Polyatomic Ions Extension Questions Answers Compatibility with Devices
 - Polyatomic Ions Extension Questions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Extension Questions Answers
 - Highlighting and Note-Taking Polyatomic Ions Extension Questions Answers
 - Interactive Elements Polyatomic Ions Extension Questions Answers
- 8. Staying Engaged with Polyatomic Ions Extension Questions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Extension Questions Answers
- 9. Balancing eBooks and Physical Books Polyatomic Ions Extension Questions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Extension Questions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ions Extension Questions Answers
 - Setting Reading Goals Polyatomic Ions Extension Questions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ions Extension Questions Answers
 - Fact-Checking eBook Content of Polyatomic Ions Extension Questions Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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