

## Lesson Quiz

# Observing Chemical Change

Write the letter of the correct answer on the line at the left.

1. \_\_\_\_ Which of the following is true about chemical reactions?  
A They are accompanied by changes in energy.  
B They form new substances with new properties.  
C both A and B  
D neither A nor B
2. \_\_\_\_ In an endothermic reaction, energy is  
A absorbed  
B released  
C converted to mass  
D synthesized
3. \_\_\_\_ Which of the following is NOT a physical property?  
A melting point  
B state of matter  
C density  
D flammability
4. \_\_\_\_ Substances formed as a result of a chemical reaction are called  
A catalysts  
B precipitates  
C products  
D reactants

If the statement is true, write *true*. If the statement is false, change the underlined word or words to make the statement true.

5. \_\_\_\_\_ In an exothermic reaction, products have more energy than reactants.
6. \_\_\_\_\_ Water boils at 100°C. This is an example of a chemical property.
7. \_\_\_\_\_ Substances that enter into a chemical reaction are called products.
8. \_\_\_\_\_ The ability to react with oxygen is an example of a chemical property.
9. \_\_\_\_\_ Another name for a chemical change is a chemical bond.
10. \_\_\_\_\_ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

# Observing Chemical Changes Workbook Answers

**Zhe Xu**



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**Learning Chemistry 8 Solution Book (Year 2023-24)** ,2024-01-02      **Learning Chemistry 7 Solution Book (Year 2023-24)** ,2024-01-02      *Scientifica* ,2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities      *Cambridge Primary Science Stage 6 Teacher's Resource Book with CD-ROM* Fiona Baxter,Liz Dilley,2014-05-22 Cambridge Primary Science is a flexible engaging course written specifically for the Cambridge Primary Science curriculum framework This Teacher s Resource for Stage 6 contains guidance on all components in the series Select activities and exercises to suit your teaching style and your learners abilities from the wide range of ideas presented Guidance includes suggestions for differentiation and assessment and supplementing your teaching with resources available online to help tailor your scheme of work according to your needs Answers to questions from the Learner s Book and Activity Book are also included The material is presented in editable format on CD ROM as well as in print to give you the opportunity to adapt it to your needs      **Learning Elementary Science Class 8 Teacher Resource Book (Academic Year 2023-24)** ,2023-05-20 Learning Elementary Science Class 8 Teacher Resource Book Academic Year 2023 24      *Microscale Chemistry* John Skinner,1997 Developing microscale chemistry experiments using small quantities of chemicals and simple equipment has been a recent initiative in the UK Microscale chemistry experiments have several advantages over conventional experiments They use small quantities of chemicals and simple equipment which reduces costs The disposal of chemicals is easier due to the small quantities Safety hazards are often reduced and many experiments can be done quickly Using plastic apparatus means glassware breakages are minimised Practical work is possible outside a laboratory Microscale Chemistry is a book of such experiments designed for use in schools and colleges and the ideas behind the experiments in it come from many sources including chemistry teachers from all around the world Current trends indicate that with the likelihood of further environmental legislation the need for microscale chemistry teaching techniques and experiments is likely to grow This book should serve as a guide in this process      **Metals and Chemical Change** D A Johnson,2007-10-31 This book looks at how molecules react and how the feasibility and outcome of chemical reactions can be predicted Beginning with an introduction to the concept of an activity series of metals Metals and Chemical Change then introduces chemical thermodynamics enthalpy entropy and free energy and applies the concept to both inorganic and organic elements A Case Study on batteries and fuel cells is also included The accompanying CD ROM includes video sequences of the reactions of metals with water acid and aqueous ions and gives the reader an opportunity to make experimental observations and predictions about chemical behaviour A comprehensive Data Book of chemical and physical constants is included along with a set of interactive self assessment questions The Molecular World series provides an integrated introduction to all branches of chemistry for both students wishing to specialise and those wishing to gain a broad understanding of chemistry and its relevance to the

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