

#### 4) Oxidation-Reduction Reactions

**Oxidation** - is a loss of electrons  
 - is an decrease in oxidation number

**Reduction** - is a gain of electrons  
 - is a decrease in oxidation number

**Oxidizing Agent** - brings about the oxidation of another species  
*↑ gets reduced*

**Reducing Agent** - brings about the reduction of another species  
*↑ causes something else to gain electrons by losing electrons itself (Is Oxidized)*  
*↑ gets oxidized*

★ OIL RIG (Oxidation Is Loss; Reduction Is Gain)  
 LEO-GER (Lose Electrons Oxidation ~ Gain Electrons Reduction)

Memorize common oxidizing pathways (reducing goes backward)

1. Metal  $\xrightarrow{\text{ox}}$  metal ion  
 $2n \rightarrow 2n^{2+}$
2. Metal  $\xrightarrow{\text{ox}}$  metallic ion  $\xrightarrow{\text{ox}}$  metallic ion  
 $\text{Cu} \rightarrow \text{Cu}^+ \rightarrow \text{Cu}^{2+}$
3. Halide ion  $\xrightarrow{\text{ox}}$  halogen  $\xrightarrow{\text{ox}}$  halite ion  $\xrightarrow{\text{ox}}$  halate ion (similar for  $\text{NO}_2 \xrightarrow{\text{ox}} \text{NO}_3^-$ )  
 $\text{Cl}^- \rightarrow \text{Cl}_2 \rightarrow \text{ClO}^- \rightarrow \text{ClO}_2^- \rightarrow \text{ClO}_3^- \rightarrow \text{ClO}_4^-$

Memorize common oxidizing / reducing agents (see bright orange sheet)

Ex.  $\text{MnO}_4^- \rightarrow \text{Mn}^{2+}$  (acidified solution)  
 $\text{MnO}_4^- \rightarrow \text{MnO}_2$  (basic solution)  
 $\text{Cr}_2\text{O}_7^{2-} \rightarrow \text{Cr}^{3+}$  (acidified solution)

===Identify the substances being oxidized and reduced in each of the following.===



MEMORIZE!

# Oxidation And Reduction Practice Answer Key

**Liying Dong**



**Oxidation And Reduction Practice Answer Key:**

## Reviewing **Oxidation And Reduction Practice Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Oxidation And Reduction Practice Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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