

Answer key-H.W(2)-Oxidation-reduction reaction

1-Multiple choice questions:

1-Oxidation involves

- a-gaining electrons; gaining oxygen*
- b-losing electrons; losing oxygen*
- c-gaining electrons; losing oxygen*
- d-losing electrons; gaining oxygen*

2- Reduction involves

- a-gaining electrons; gaining oxygen*
- b-losing electrons; losing oxygen*
- c-gaining electrons; losing oxygen*
- d-losing electrons; gaining oxygen*

3-In the reaction



- a-Reduced*
- b- Synthesized*
- c-Oxidized*

4- In the reaction $\text{Zn} + \text{H}_2\text{O} \rightarrow \text{ZnO}_2 + \text{H}_2$

which element, is oxidized?

- a-Zinc*
- b- Hydrogen*
- c- Oxygen*

5- Which substance is reduced in the following reaction?



- a-Li*
- b-Na*
- c-O*
- d-H*

6-Which of the following is the balanced half -reaction for the oxidation of Cu to Cu^{+2} ?


- a-Cu $\rightarrow \text{Cu}^{+2}$*
- b-Cu + 2 e⁻ $\rightarrow \text{Cu}^{+2}$*
- c-Cu $\rightarrow \text{Cu}^{+2} + 2 \text{e}^-$*

7-Another name for a "oxidation-reduction" reaction is

- a-chemical reaction*
- b-neutralization reaction*
- c-redox reaction*

Remember

Redox (Oxidation and Reduction)



Oxidation	Reduction
Gain oxygen ↑	Loss oxygen ↓
Gain oxidation Number ↑	Loss oxidation Number ↓
Loss hydrogen ↓	Gain hydrogen ↑
Loss electron ↓	Gain electron ↑

Oxidation Reduction Chemquest Answer Key

Robert R. Redfield



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,1974 **Reactions Introduction** , **Oxidation - Reduction Equation** George Howard Satterfield,1936

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