

Lesson Quiz

Observing Chemical Change

Write the letter of the correct answer on the line at the left.

1. ____ Which of the following is true about chemical reactions?
A They are accompanied by changes in energy.
B They form new substances with new properties.
C both A and B
D neither A nor B
2. ____ In an endothermic reaction, energy is
A absorbed
B released
C converted to mass
D synthesized
3. ____ Which of the following is NOT a physical property?
A melting point
B state of matter
C density
D flammability
4. ____ Substances formed as a result of a chemical reaction are called
A catalysts
B precipitates
C products
D reactants

If the statement is true, write *true*. If the statement is false, change the underlined word or words to make the statement true.

5. _____ In an exothermic reaction, products have more energy than reactants.
6. _____ Water boils at 100°C. This is an example of a chemical property.
7. _____ Substances that enter into a chemical reaction are called products.
8. _____ The ability to react with oxygen is an example of a chemical property.
9. _____ Another name for a chemical change is a chemical bond.
10. _____ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

Pearson Observing Chemical Change

Haddy James



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Prentice Hall Science Explorer: Chemical Interactions Pearson/Prentice Hall, Michael J. Padilla, 2004-10 **A system of surgery, tr. and accompanied with notes and observations b J.F. South** Maximilian Joseph Chelius, 1847

A treatise on the chemical history and medical powers of some of the most celebrated mineral waters; with practical remarks on the aqueous regimen. To which are added, observations on the use of cold and warm bathing William SAUNDERS (M.D.), 1800 A Treatise on the Chemical History and Medical Powers of Some of the Most Celebrated Mineral Waters William Saunders, 1800

Focus on Physical Science California Edition Michael J. Padilla, 2007 *Proceedings of the Royal Society of London* Royal Society (Great Britain), 1896 Prentice Hall Chemistry Antony C. Wilbraham, 2008

Observations on Some of the General Principles and on the Particular Nature and Treatment of the Different Species of Inflammation Haddy James, 1821 **Index to the Proceedings of the Royal Society of London (old Series)** , 1913 Abstracts of the Papers Printed in the Philosophical Transactions of the Royal Society of London Royal Society (Great Britain), 1913 **Proceedings of the Royal Society of London** , 1913 Index to the Proceedings of the Royal Society of London , 1913 **Principles of Surgery** John Pearson, 1832 *On Being*

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